

# PALEOENVIRONMENTAL INTERPRETATION OF EVAPORITIC FACIES

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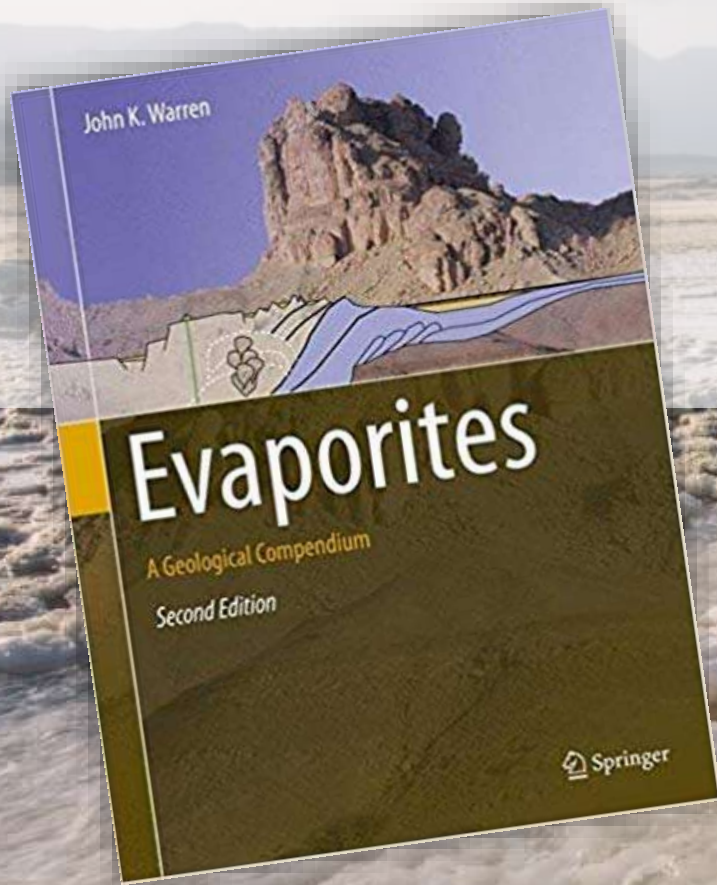


***Saltgiant training school- Salamanca, November 2019***

# Outline

1. Definition of evaporites
2. Evaporitic minerals and brine chemistry
3. Modern evaporites
4. Ancient evaporites (gypsum) and associated sediments (shales)  
*Facies and depositional environment*

# Evaporites: a definition



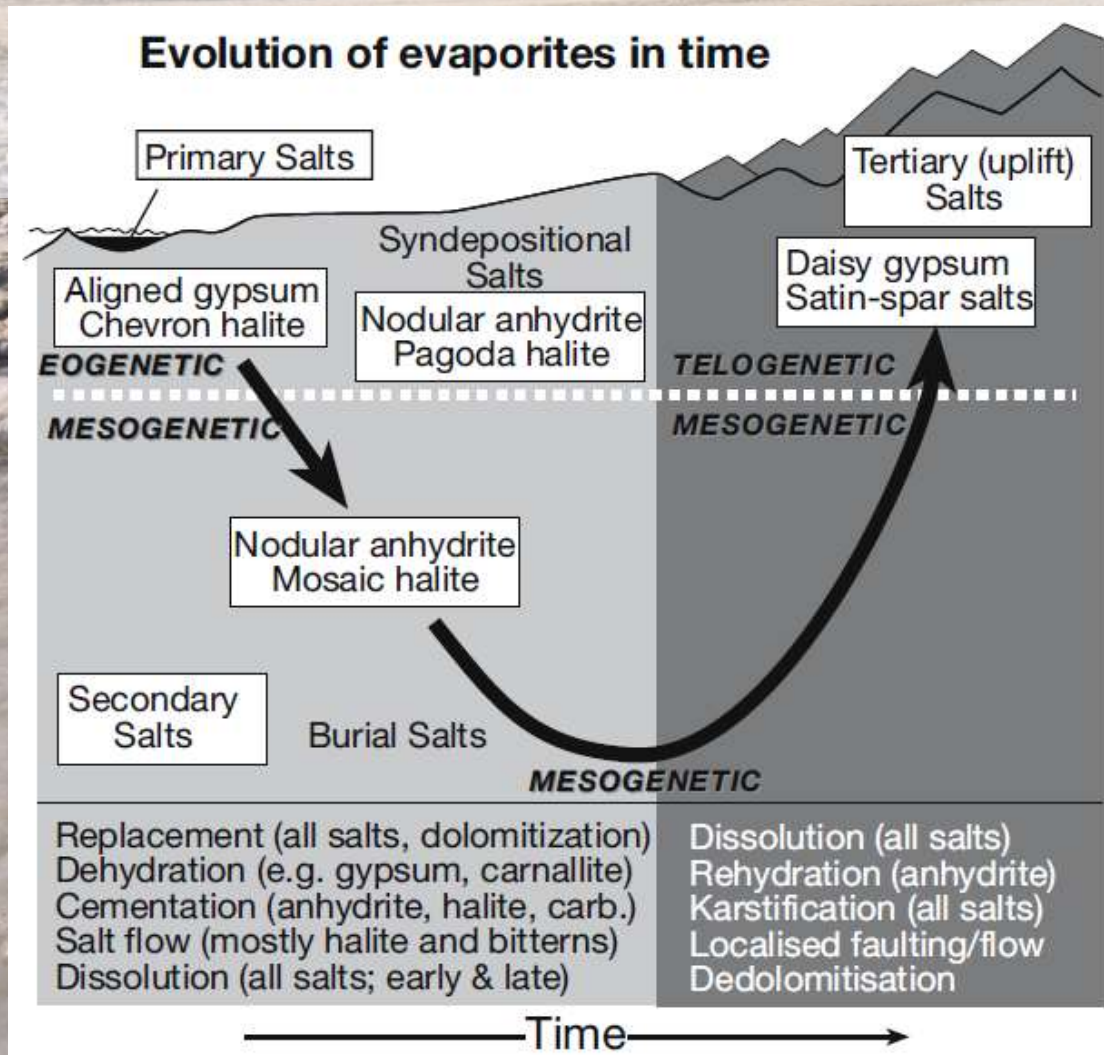
*A salt rock that was originally precipitated from a saturated surface or nearsurface brine concentrated by solar evaporation.”  
(Warren, 2006)*

*Salt deposits: salt rocks formed by other processes than evaporation*

# Evaporites: a definition

**Primary evaporites:** Texture and lithology are a direct evidence of the hydrology of the basin at time of precipitation/accumulation.

**Secondary evaporites:** during burial and exhumation.



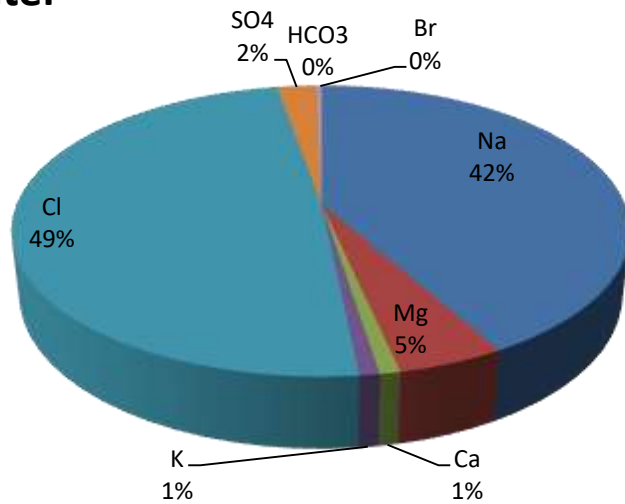
# Brine chemistry and evaporitic minerals

The mineralogy of evaporites strongly depends from the type and the chemistry of their mother brines (seawater, continental, hydrothermal, or a mixture of them)

Modified from Warren, 2006

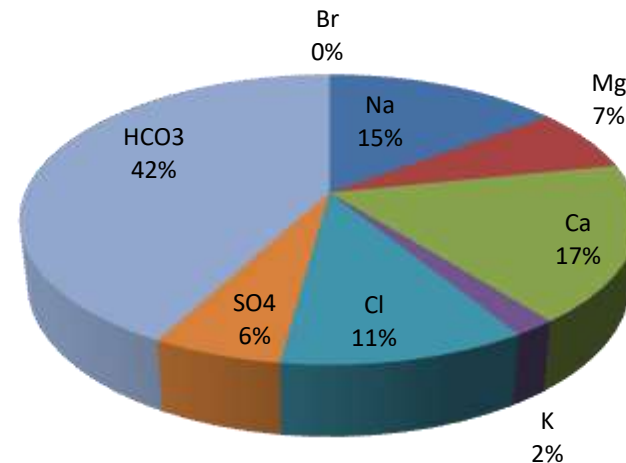
Locality	Salinity (‰)	Na	Mg	Ca	K	Cl	SO <sub>4</sub>	HCO <sub>3</sub>	Br
Seawater (average)	35	468	53	10	10	546	28	2	1
River water (average)	0.11	0.31	0.15	0.37	0.04	0.23	0.12	0.89	-

## Seawater



Marine

## River

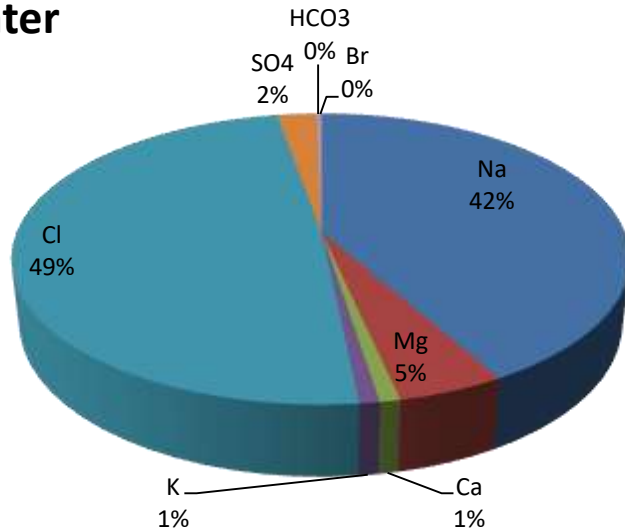


Continental

# Main evaporitic minerals

- 1) Evaporitic alkaline carbonates (aragonite, dolomite, low-Mg calcite, and high-Mg calcite)
- 2) Evaporite salts (**gypsum**, anhydrite, **halite**, sylvite, kainite, carnallite)

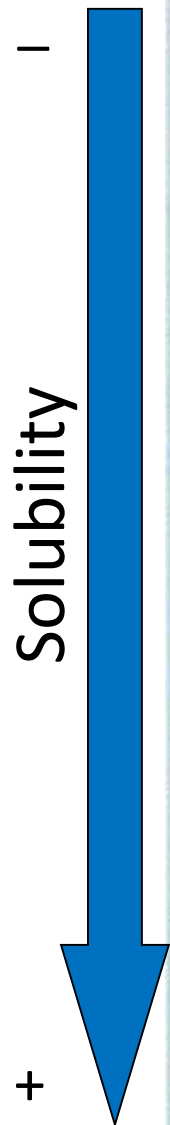
Seawater



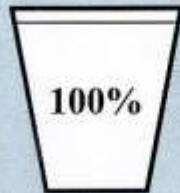
Marine

Mineral	Formula	Mineral	Formula
Anhydrite	$\text{CaSO}_4$	Leonhardite	$\text{MgSO}_4 \cdot 4\text{H}_2\text{O}$
Antarctite	$\text{CaCl}_2 \cdot 6\text{H}_2\text{O}$	Leonite	$\text{MgSO}_4 \cdot \text{K}_2\text{SO}_4 \cdot 4\text{H}_2\text{O}$
Aphthitalite (glaserite)	$\text{K}_2\text{SO}_4 \cdot \text{Na}_2\text{K}_2\text{SO}_4$	Loewite	$2\text{MgSO}_4 \cdot 2\text{Na}_2\text{SO}_4 \cdot 5\text{H}_2\text{O}$
Aragonite**	$\text{CaCO}_3$	Mg-calcite**	$(\text{Mg}, \text{Ca})_2\text{CO}_3$
Bassanite	$\text{CaSO}_4 \cdot 1/2\text{H}_2\text{O}$	Magnesite**	$\text{MgCO}_3$
Bischofite	$\text{MgCl}_2 \cdot 6\text{H}_2\text{O}$	Meyerhoffite	$\text{Ca}_3\text{B}_3\text{O}_{11} \cdot 7\text{H}_2\text{O}$
Bloedite (astrakanite)	$\text{Na}_2\text{SO}_4 \cdot \text{MgSO}_4 \cdot 4\text{H}_2\text{O}$	Mirabilite	$\text{Na}_2\text{SO}_4 \cdot 10\text{H}_2\text{O}$
Borax (tincal)	$\text{Na}_2\text{B}_4\text{O}_7 \cdot 10\text{H}_2\text{O}$	Nahcolite	$\text{NaHCO}_3$
Boracite	$\text{Mg}_2\text{B}_2\text{O}_7 \cdot \text{Cl}$	Natron	$\text{Na}_2\text{CO}_3 \cdot 10\text{H}_2\text{O}$
Burkeite	$\text{Na}_2\text{CO}_3 \cdot 2\text{Na}_2\text{SO}_4$	Nitrate (soda nitre)	$\text{NaNO}_3$
Calcite**	$\text{CaCO}_3$	Nitre (salt petre)	$\text{KNO}_3$
Carnallite	$\text{MgCl}_2 \cdot \text{KCl} \cdot 6\text{H}_2\text{O}$	Pentahydrate	$\text{MgSO}_4 \cdot 5\text{H}_2\text{O}$
Colemanite	$\text{Ca}_2\text{B}_6\text{O}_{11} \cdot 5\text{H}_2\text{O}$	Pisronite	$\text{CaCO}_3 \cdot \text{Na}_2\text{CO}_3 \cdot 2\text{H}_2\text{O}$
Darapskite	$\text{NaSO}_3 \cdot \text{NaNO}_2 \cdot \text{H}_2\text{O}$	Polyhalite	$2\text{CaSO}_4 \cdot \text{MgSO}_4 \cdot \text{K}_2\text{SO}_4 \cdot \text{H}_2\text{O}$
Dolomite**	$\text{Ca}_{1-x}\text{Mg}_x(\text{CO}_3)_2$	Proberite	$\text{NaCaB}_3\text{O}_7 \cdot 5\text{H}_2\text{O}$
Epsomite	$\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$	Pricite (pandermite)	$\text{CaB}_3\text{O}_7 \cdot 7\text{H}_2\text{O}$
Ferrosulphate	$3\text{NaSO}_4 \cdot \text{Fe}_2(\text{SO}_4)_3 \cdot 6\text{H}_2\text{O}$	Rinnite	$\text{FeCl}_2 \cdot \text{NaCl} \cdot 3\text{KCl}$
Gaylussite	$\text{CaCO}_3 \cdot \text{Na}_2\text{CO}_3 \cdot 5\text{H}_2\text{O}$	Sanderite	$\text{MgSO}_4 \cdot 2\text{H}_2\text{O}$
Glauberite	$\text{CaSO}_4 \cdot \text{Na}_2\text{SO}_4$	Schoenite (picromerite)	$\text{MgSO}_4 \cdot \text{K}_2\text{SO}_4 \cdot 6\text{H}_2\text{O}$
Gypsum	$\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$	Shorite	$2\text{CaCO}_3 \cdot \text{Na}_2\text{CO}_3$
Halite	$\text{NaCl}$	Sylvite	$\text{KCl}$
Hanksite	$9\text{Na}_2\text{SO}_4 \cdot 2\text{Na}_2\text{CO}_3 \cdot \text{KCl}$	Syngenite	$\text{CaSO}_4 \cdot \text{K}_2\text{SO}_4 \cdot \text{H}_2\text{O}$
Hexahydrate	$\text{MgSO}_4 \cdot 6\text{H}_2\text{O}$	Tachyhydrite	$\text{CaCl}_2 \cdot 2\text{MgCl}_2 \cdot 12\text{H}_2\text{O}$
Howite	$\text{H}_2\text{Ca}_2\text{SiB}_2\text{O}_{11}$	Thernadite	$\text{Na}_2\text{SO}_4$
Ikate**	$\text{CaCO}_3 \cdot 6\text{H}_2\text{O}$	Thermonatrite	$\text{NaCO}_3 \cdot \text{H}_2\text{O}$
Inyoite	$\text{Ca}_2\text{B}_6\text{O}_{11} \cdot 13\text{H}_2\text{O}$	Tincalconite	$\text{Na}_2\text{B}_4\text{O}_7 \cdot 5\text{H}_2\text{O}$
Kainite	$4\text{MgSO}_4 \cdot 4\text{KCl} \cdot 11\text{H}_2\text{O}$	Trona	$\text{NaHCO}_3 \cdot \text{Na}_2\text{CO}_3$
Kernite	$\text{Na}_2\text{B}_4\text{O}_7 \cdot 4\text{H}_2\text{O}$	Tychite	$2\text{MgCO}_3 \cdot 2\text{NaCO}_3 \cdot \text{Na}_2\text{SO}_4$
Kieserite	$\text{MgSO}_4 \cdot \text{H}_2\text{O}$	Ulexite	$\text{NaCaB}_3\text{O}_7 \cdot 5\text{H}_2\text{O}$
Langbeinite	$2\text{MgSO}_4 \cdot \text{K}_2\text{SO}_4$	Vanthoffite	$\text{MgSO}_4 \cdot 3\text{Na}_2\text{SO}_4$

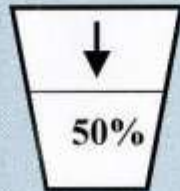
# Marine evaporites



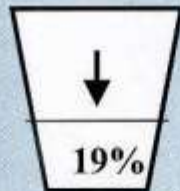
## Usiglio's Experiment (1849): Evaporation of Sea Water



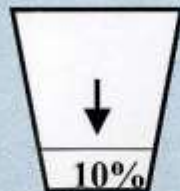
Salinity  
35 - 37 ‰



60 - 80 ‰  
Carbonate ( $\text{CaCO}_3$ )

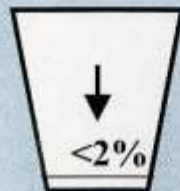


185 - 200 ‰  
Gypsum ( $\text{CaSO}_4 \times 2\text{H}_2\text{O}$ )



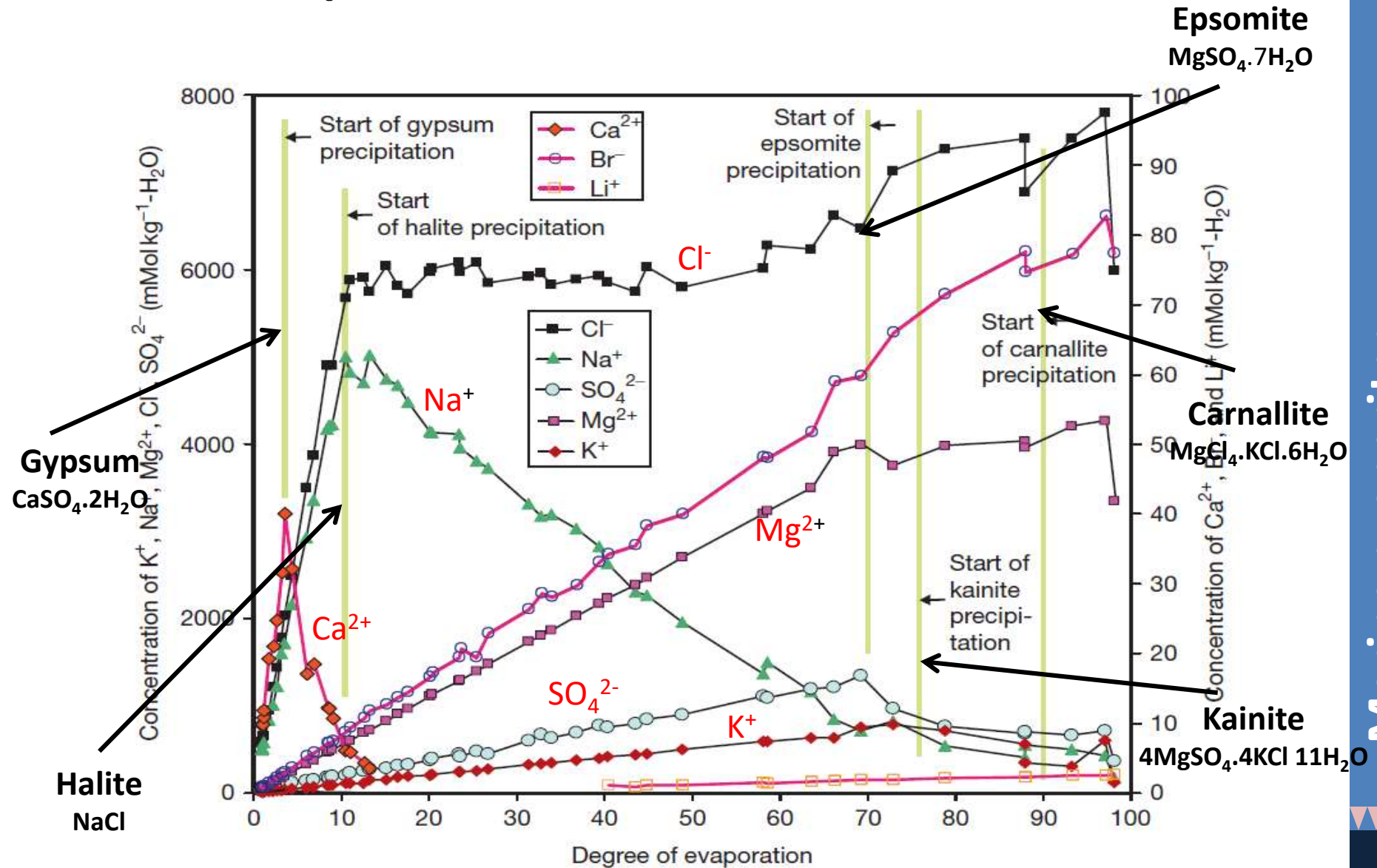
350 ‰

Halite ( $\text{NaCl}$ )



Bitter Salts  
( $\text{Mg-K-Cl-SO}_4$  Salt)

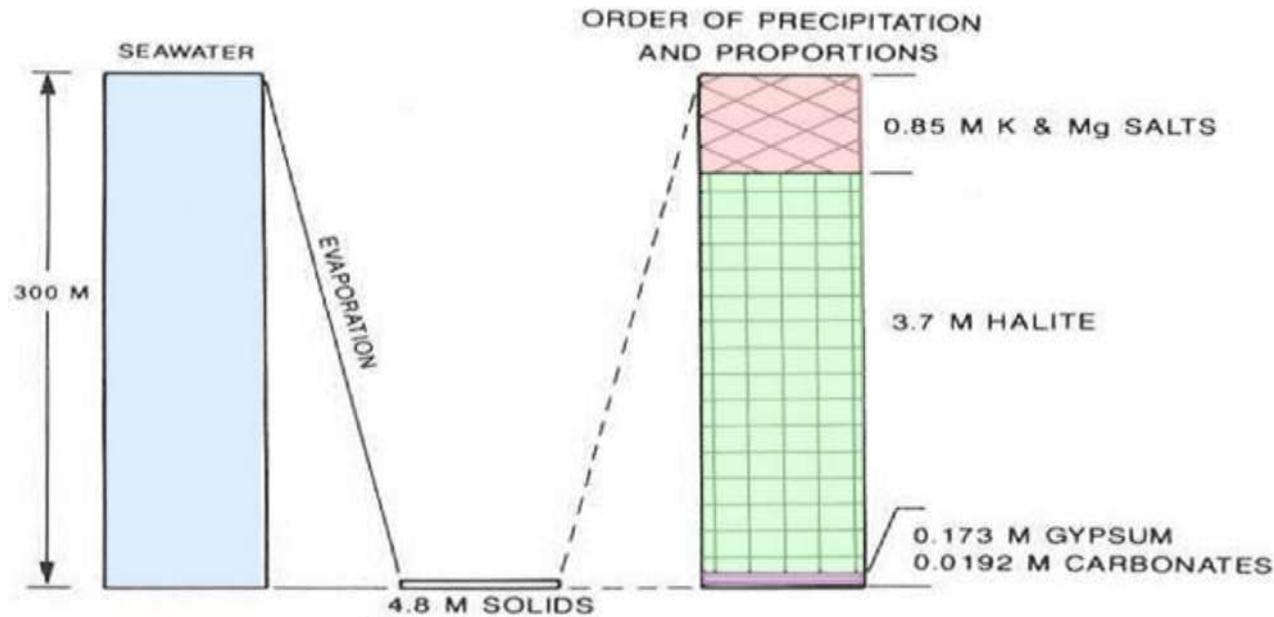
# Marine evaporites



from Babel and Schreiber, 2014



# Marine evaporites: rate of deposition



SEDIMENT TYPE	ENVIRONMENT	RATE OF DEPOSITION
Sulphates and carbonates	Sabkha (supratidal)	1m/1000yrs + progradation 1km/1000yrs
Sulphates	Shallow subaqueous	<b>1-40 m/1000 yrs</b>
Sulphates	Deep subaqueous	<b>1-2 m /1000 yrs</b>
Halite	Shallow subaqueous	<b>10-100 m/1000 yrs</b>
Halite	Deep subaqueous	<b>1-10 m /1000 yrs</b>
Reefal carbonates	Shallow water	1-3 m/ 1000 yrs
Non-reefal carbonates	Shelf	1-3 cm/ 1000 yrs
Pelagic carbonates	Deep sea	0.05-0.1 cm/ 1000 yrs

# Modern evaporites

- *Observations of modern evaporites form the core of our understanding of evaporite deposition in the geological past!*
- *Mostly shallow water (salt works) and subaerial (sabhka) environments. No modern analogues for (inferred) deep water settings*



# **Modern salt works** *Trapani (Sicily)*

**Gypsum "crystallizer" pond**

**Salinity > 110 ‰**



**0.5-1 cm year**

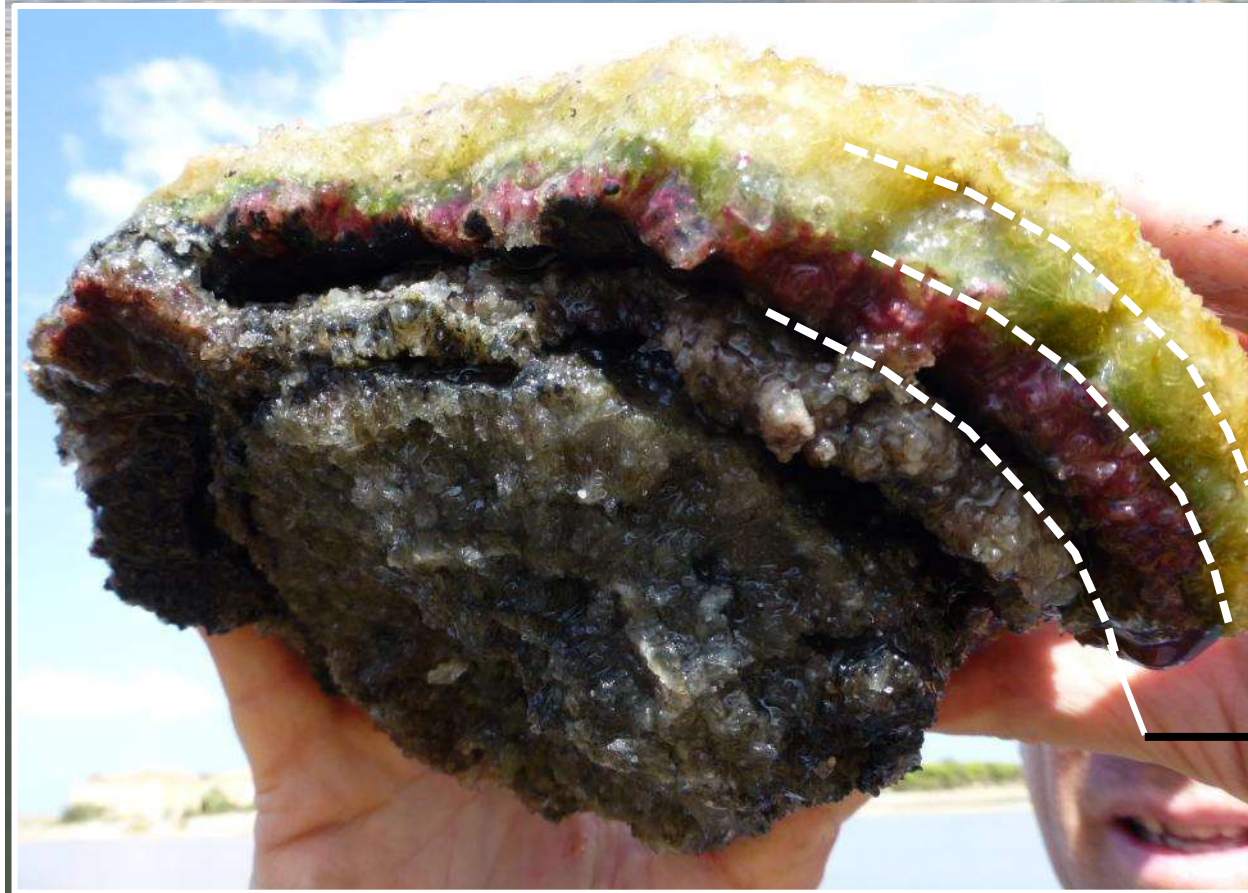
5 cm

~8 years

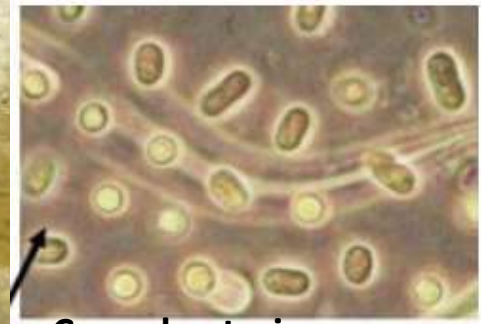
# Micro-organisms in modern gypsum

*Modern saltworks*

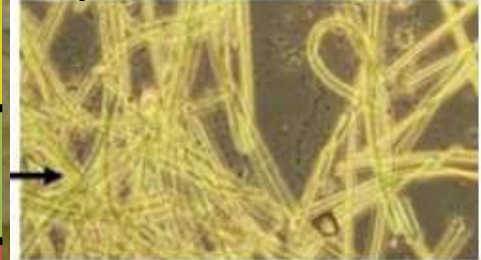
*Gypsum layering*



- Brown
- Green
- Purple
- Black



Cyanobacteria



Purple bacteria

SRB

# Modern salt works *Trapani (Sicily)*

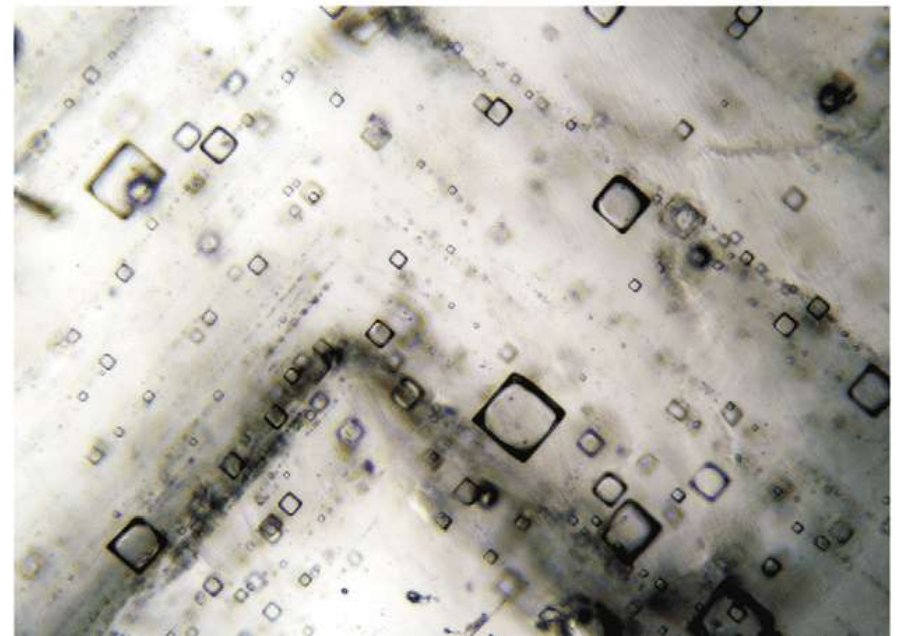
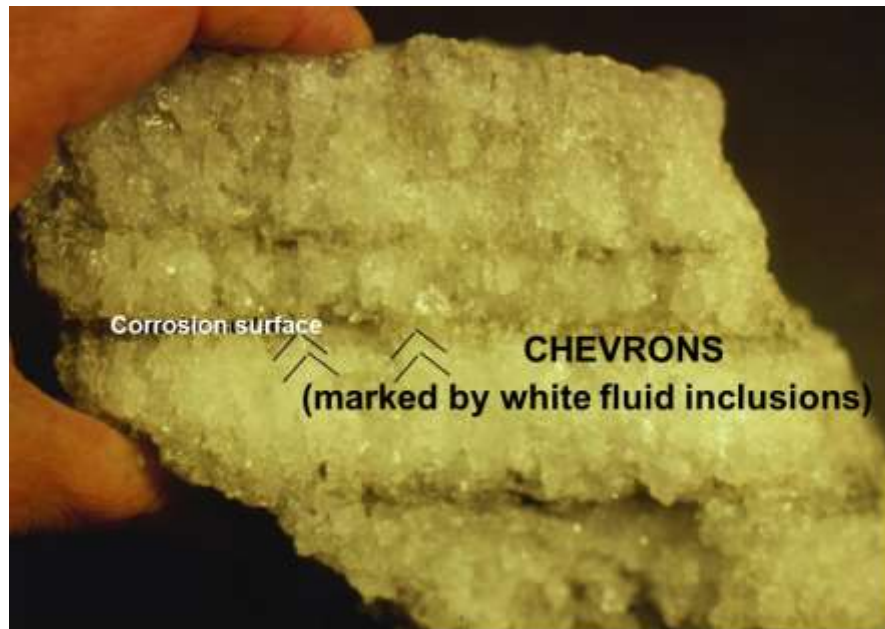
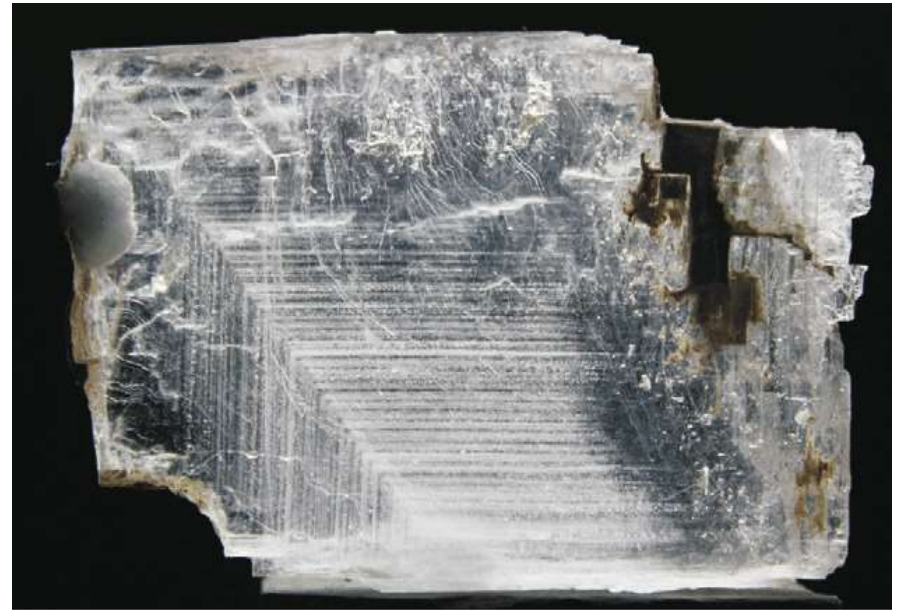
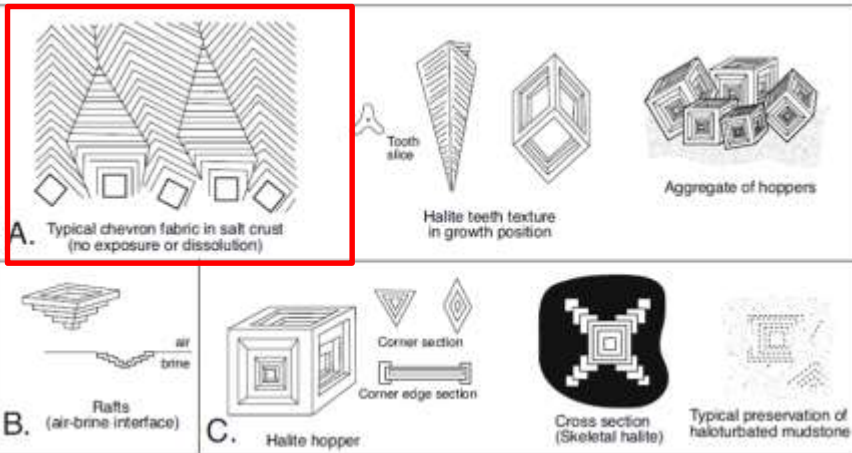
*Halite "crystallizer" pond*

*Salinity > 350 ‰*



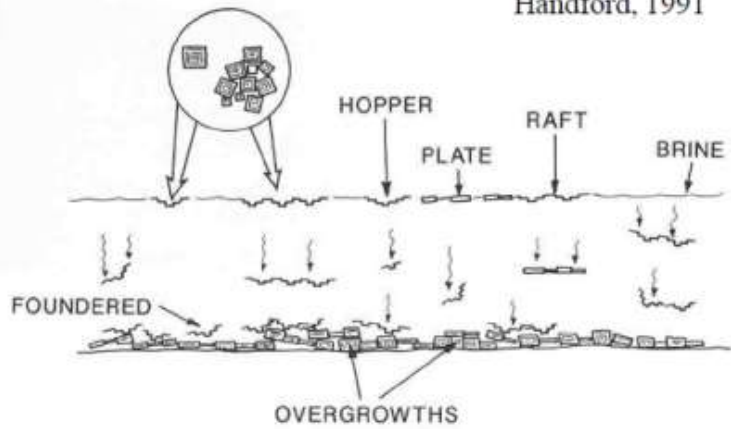
# Halite chevrons

Babel and Schreiber, 2014



# Halite rafts

Handford, 1991



a



b



d



c



e

Modern evaporites

Filippi et al. 2011

# Deep water halite (Ded Sea)





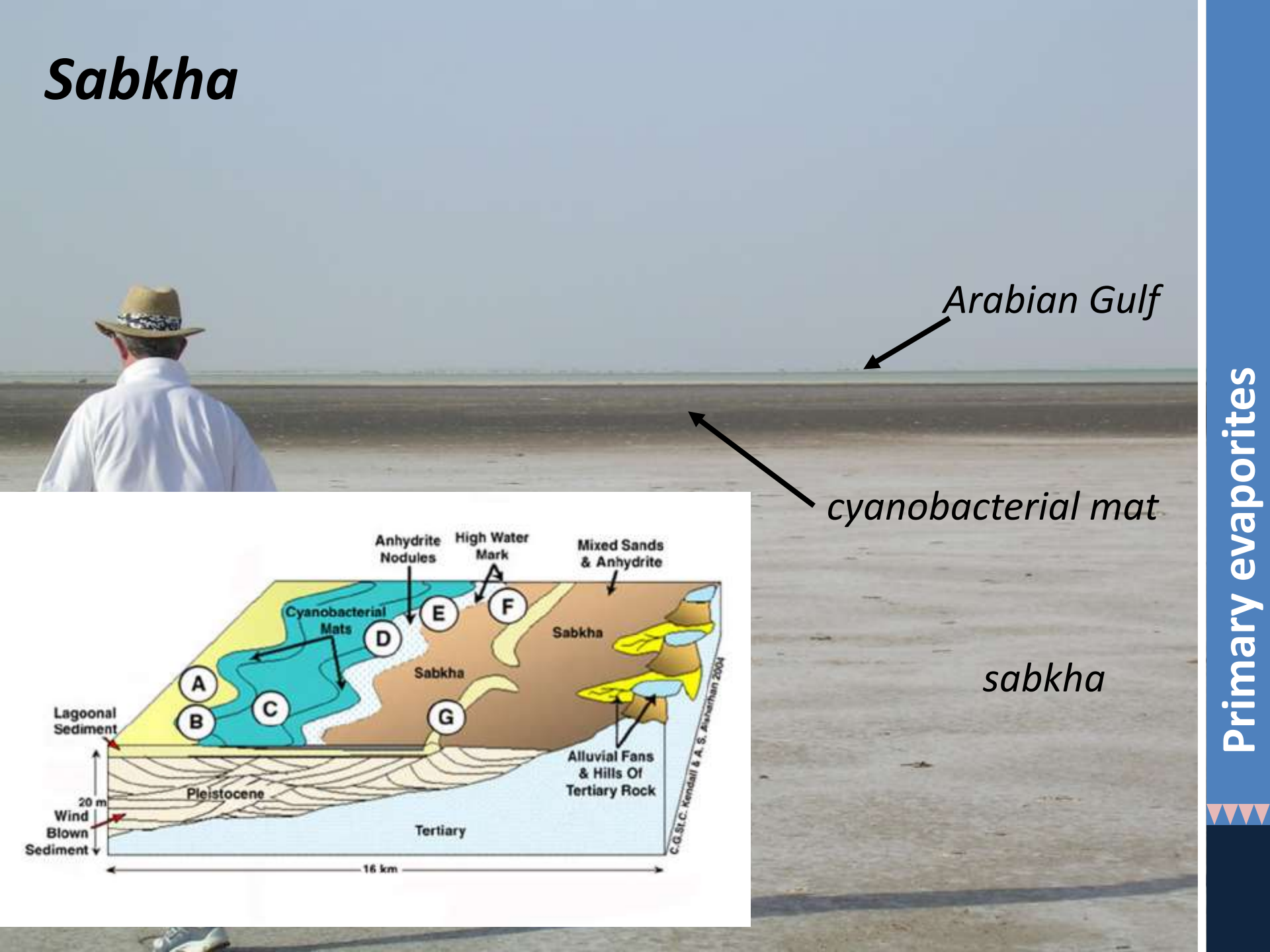
# Sabkha



Arabian Gulf

*“ the sabkha is a groundwater driven sedimentary system indicated by syndepositional intrasediment capillary evaporites, which precipitate in both marine and continental saline mudflats” (Warren, 2006)*

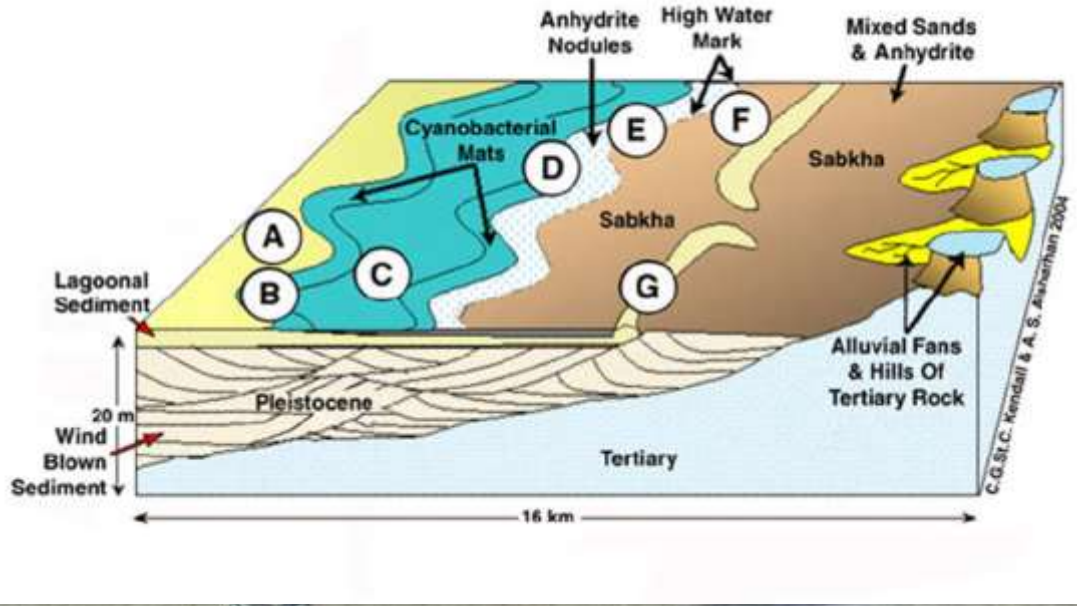
# Sabkha



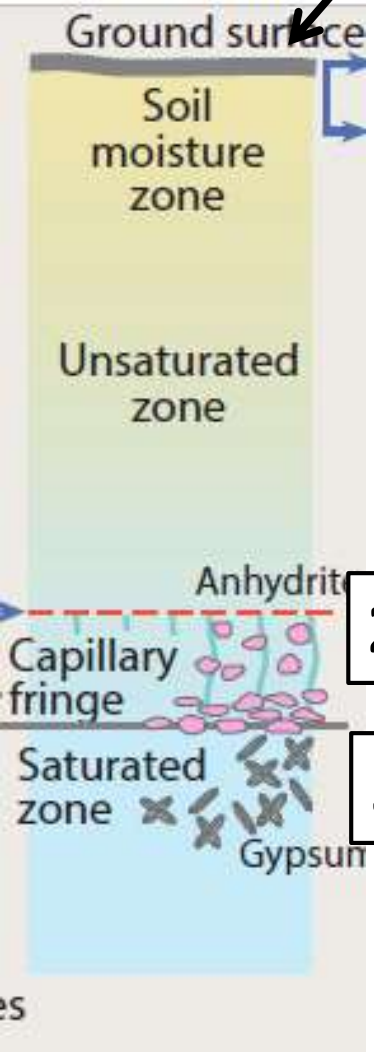
Arabian Gulf

cyanobacterial mat

sabkha



# Sabkha



Cyanobacterial mat



Anhydrite

Entherolitic anhydrite




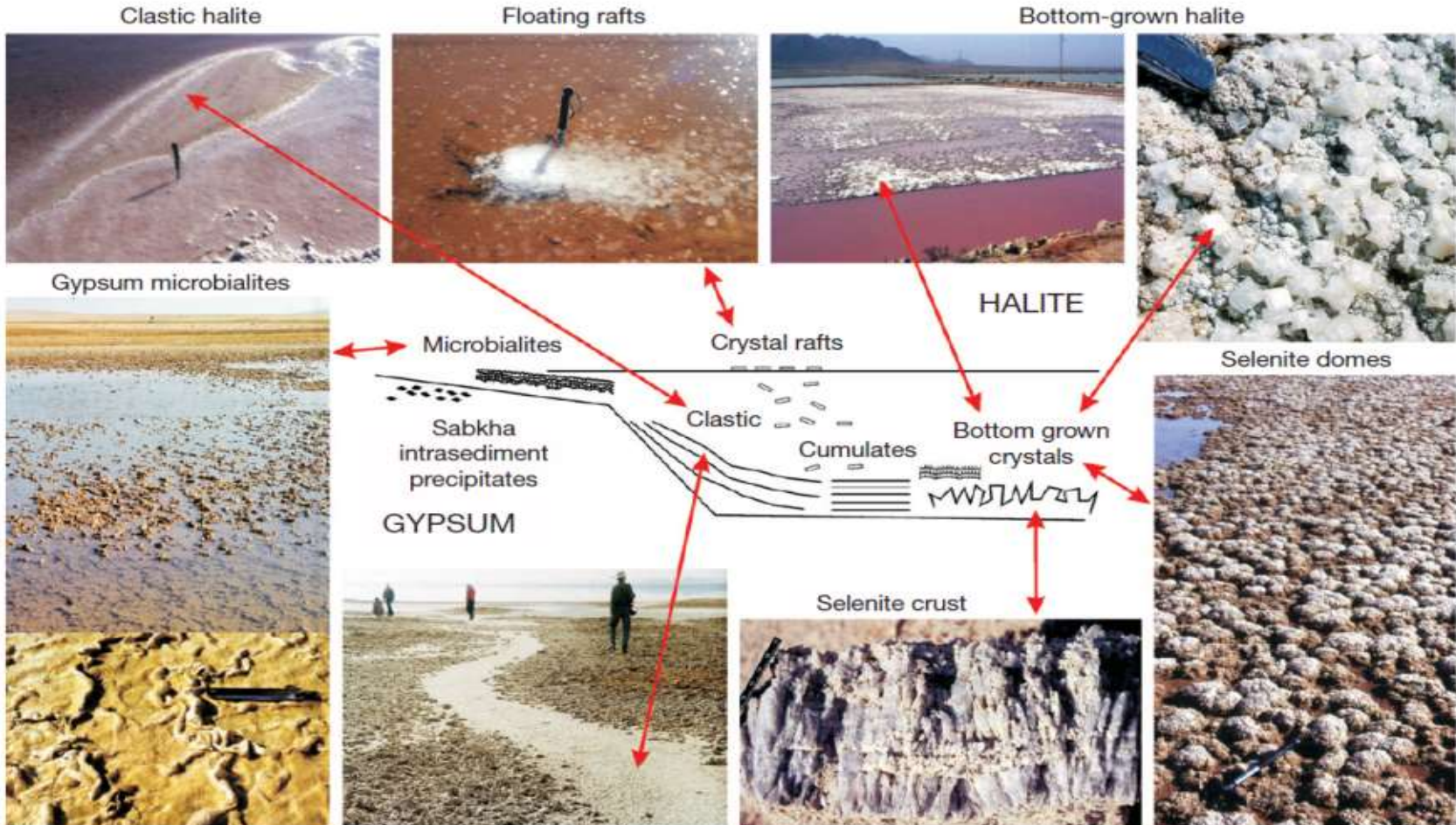
Displacive gypsum

From Warren, 2017

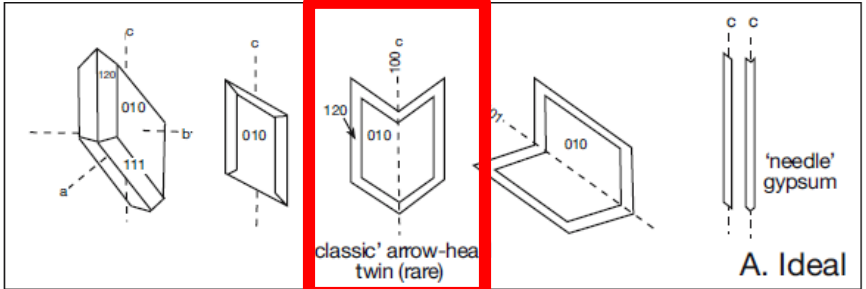
# Where evaporite form?

- at the brine/air interface
- within the brine column, particularly at the pycnocline
- on the floor of the evaporite basin
- in brine-soaked sediments (Sabkas)


**Cumulate deposits**  
**Bottom grown crystals**



# Bottom grown gypsum



# Mottura's Rule

(arrow-head, swallow's tail)



Modern evaporites

Eraclea Minoa, Sicily

# Gypsum - lenticular

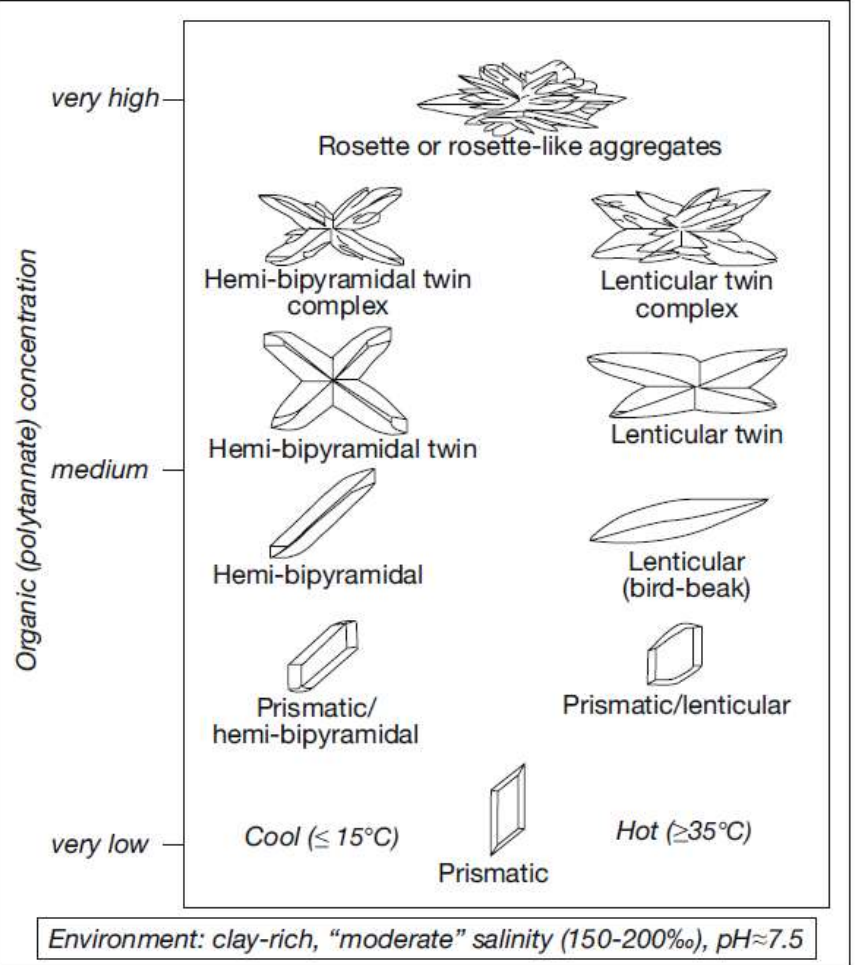
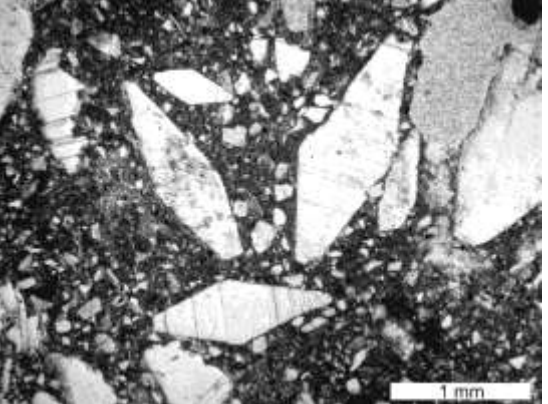
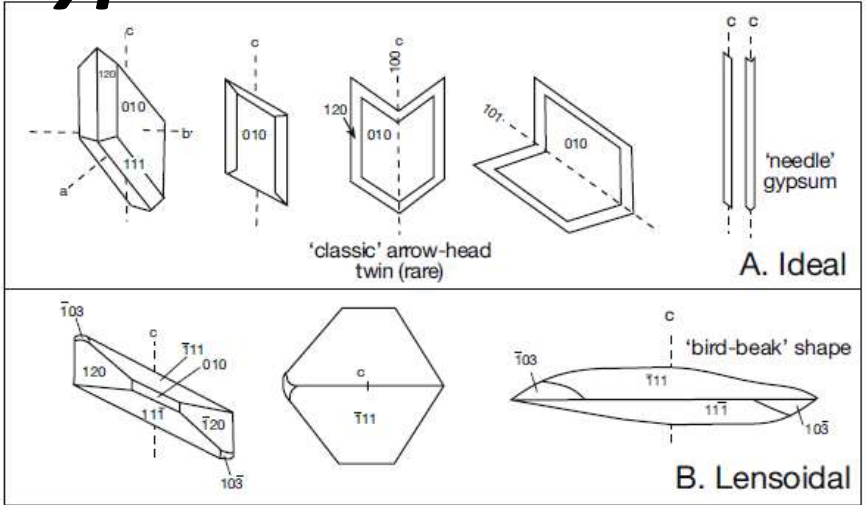


Figure 1.17. Laboratory based changes in gypsum growing in a muddy (bentonite) sediment matrix at moderate salinity and neutral pH in environments characterized by increasing temperatures and dissolved polytannate concentrations. Polytannate is considered to be an analog for terrestrial humic material. c-axis vertical in all examples (after Cody and Cody, 1988).

From Warren, 2006

# Models of marine evaporite formation

Marine marginal basin

*Crucial element*

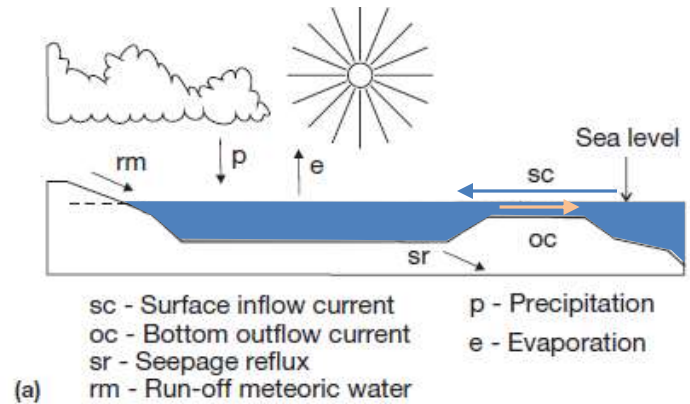


Negative hydrological budget

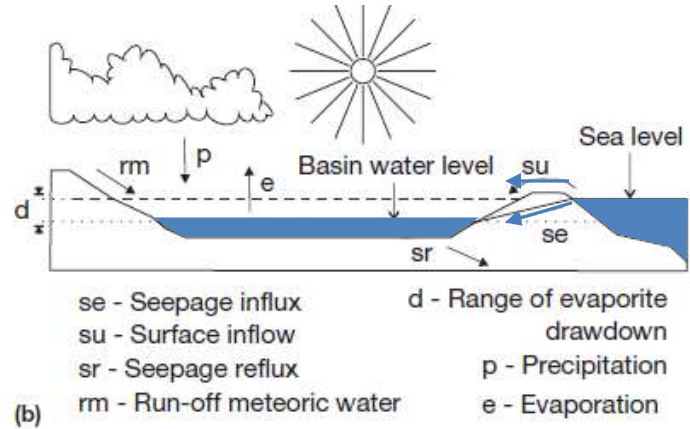
**Key factors:**

- Marginal position
- Seal or topographic barrier
- Restriction with episodically re-connection with the sea/ocean
- Climate control (aridity promoting evaporation)

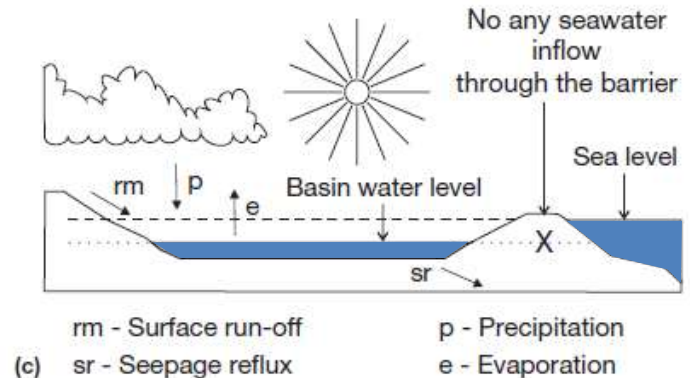
## Lagoon basin



## Salina basin

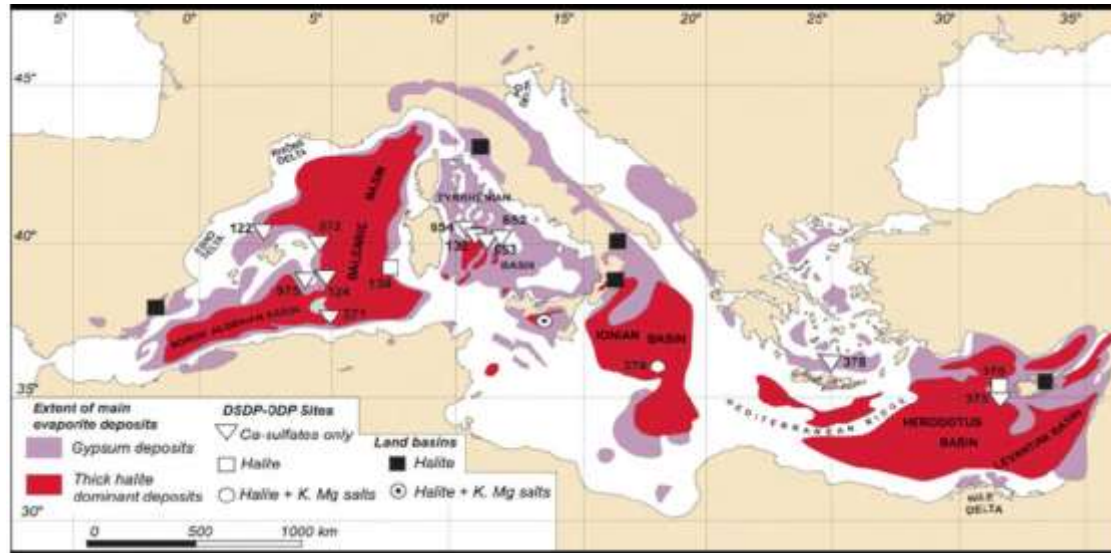


## Salina lake



from Babel and Schreiber, 2014

# Ancient evaporites (MSC)



*Rouchy and Caruso 2006*

Interpretation of ancient marine evaporite deposits is difficult!

- No (few) clues to estimate water depth
- No modern analogues for very large and (relatively) deep basins (Salt giants)

This is particularly true for the wide array of Messinian evaporites facies (mostly gypsum and halite) of the Mediterranean!



# Gypsum

## Bottom-grown gypsum

- Massive selenite
- Banded selenite
- Branching selenite («Supercones»)

## Gypsum cumulate

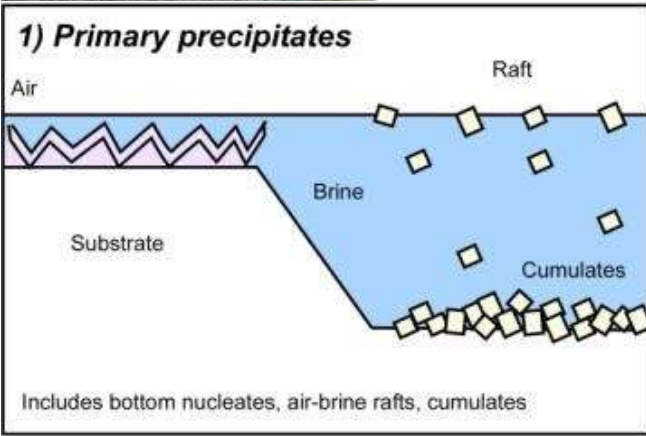
- Laminar gypsum

## Clastic gypsum



# Massive selenite

Vena del Gesso



# *Giant gypsum selenite*

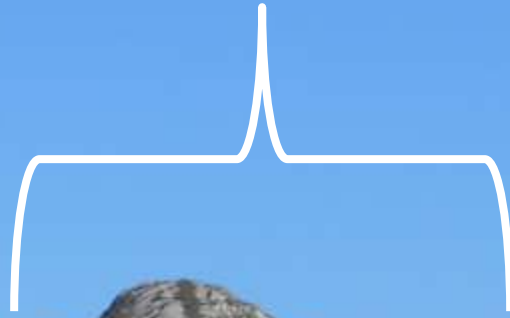
Middle Miocene (Poland)

Up to 3.5 m high



Ancient evaporites

# Upper Gypsum, Monte Palco (Sicily)



Ancient evaporites

# Nucleation cones



Sorbas



Cuevas de Sorbas

# Banded selenite



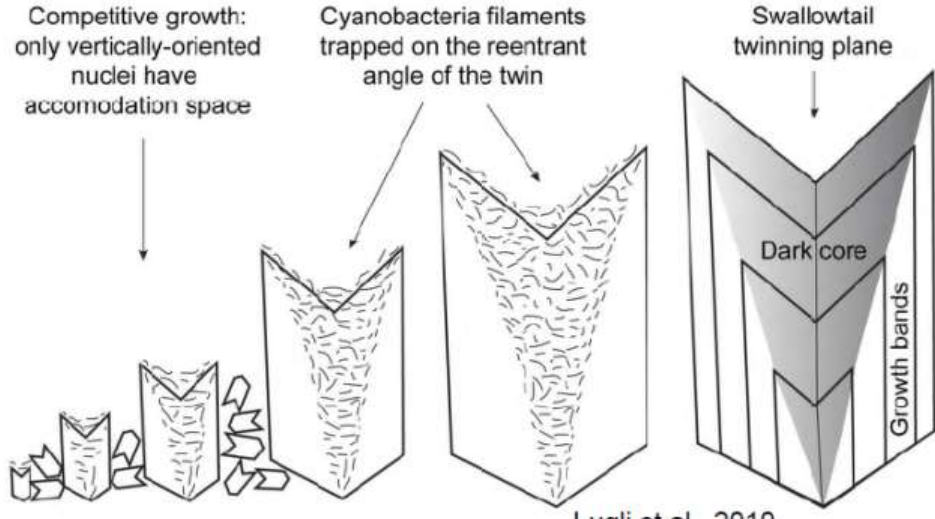
Vena del Gesso



Piedmont Basin

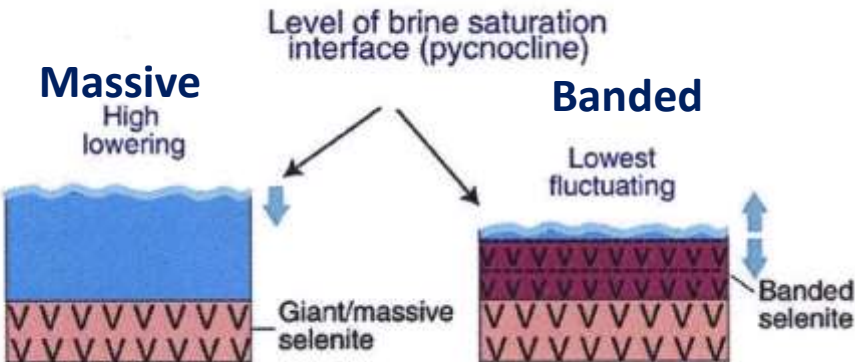


# Vertical growth of crystals in massive and banded selenite



Climatic precession cycle

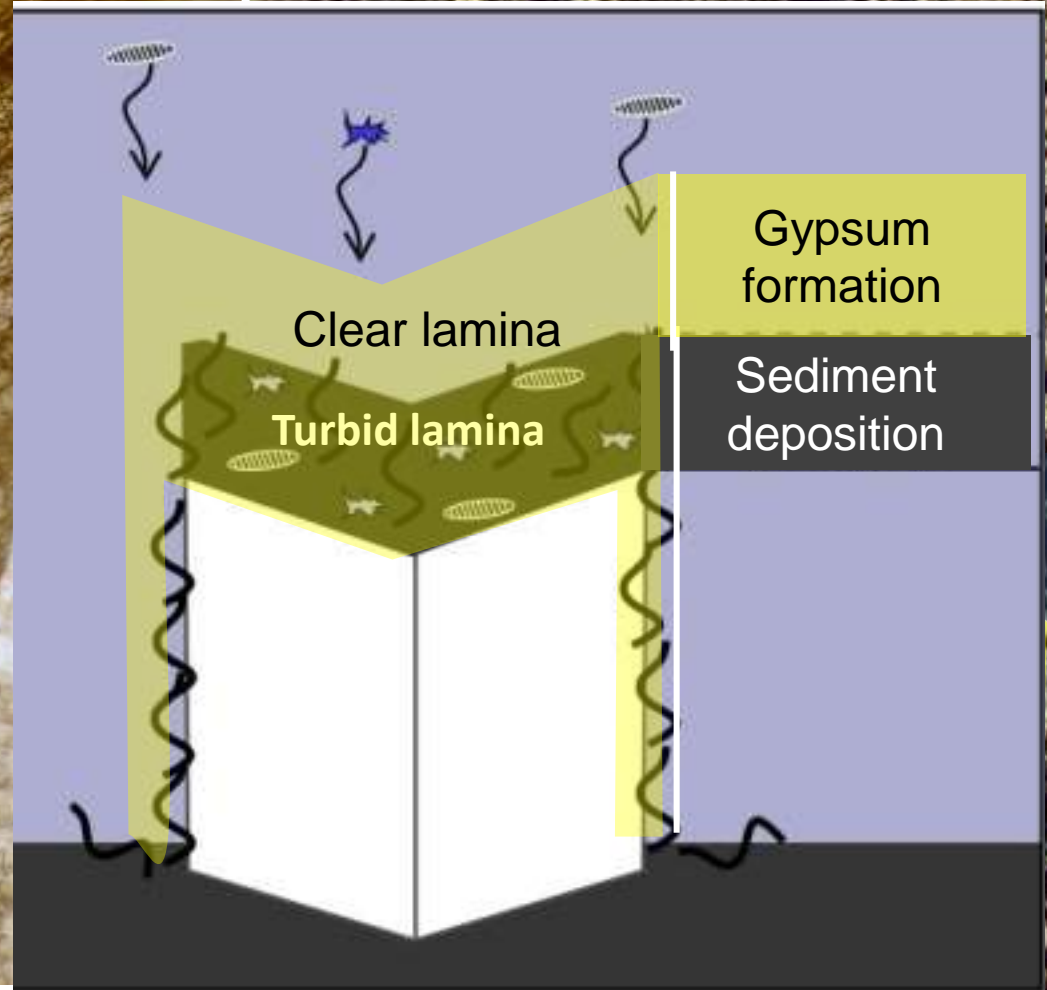
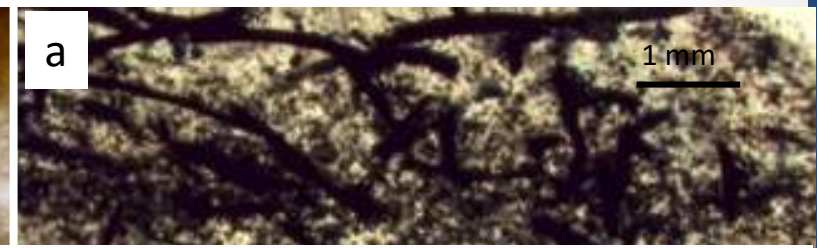
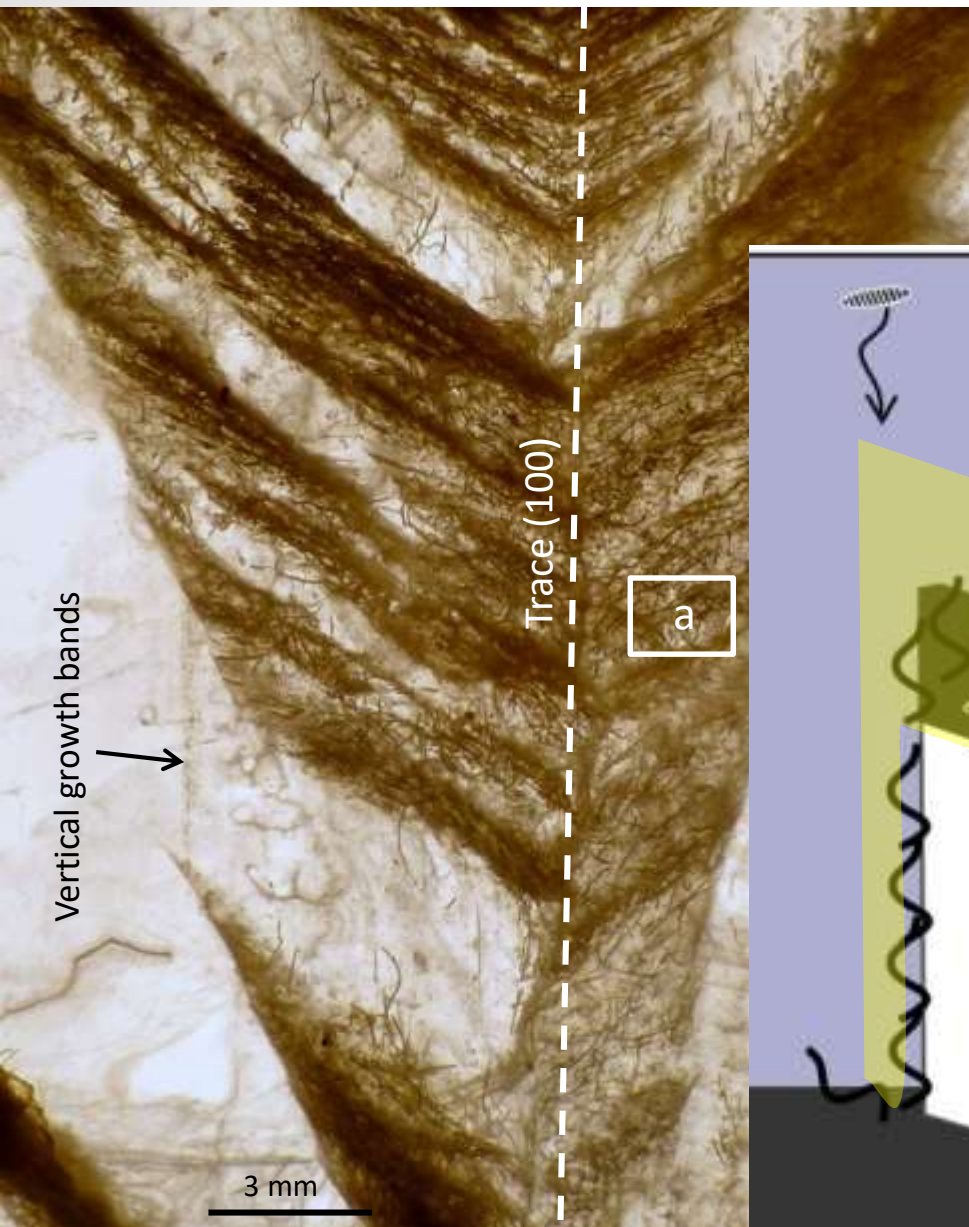
ARID



Lugli et al., 2010



# Microbiological content of massive and banded selenite



Ancient evaporites

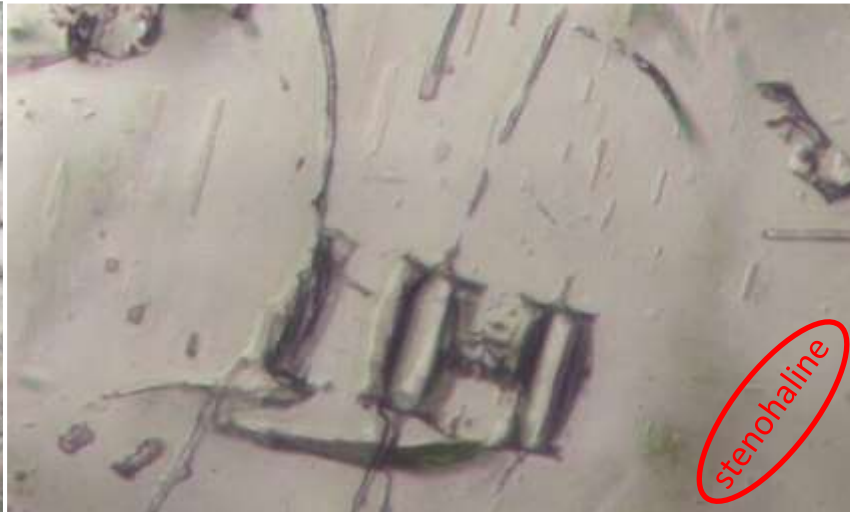


# Planktic components (only in the reentrant angle)



*Trigonium* sp.

300



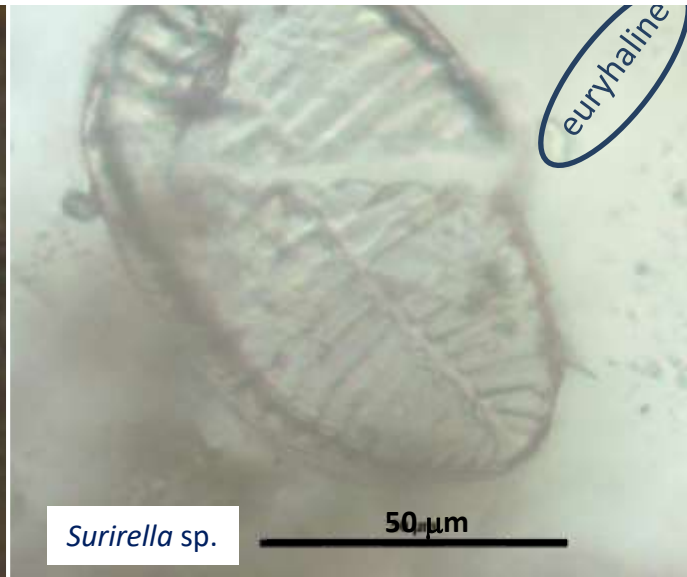
40  $\mu$ m

Mixing of steno- and euryhaline forms  
**Fresh water inflows in a marine basin!**



5  $\mu$ m

*Coscinodiscus* sp.



*Surirella* sp.

50  $\mu$ m

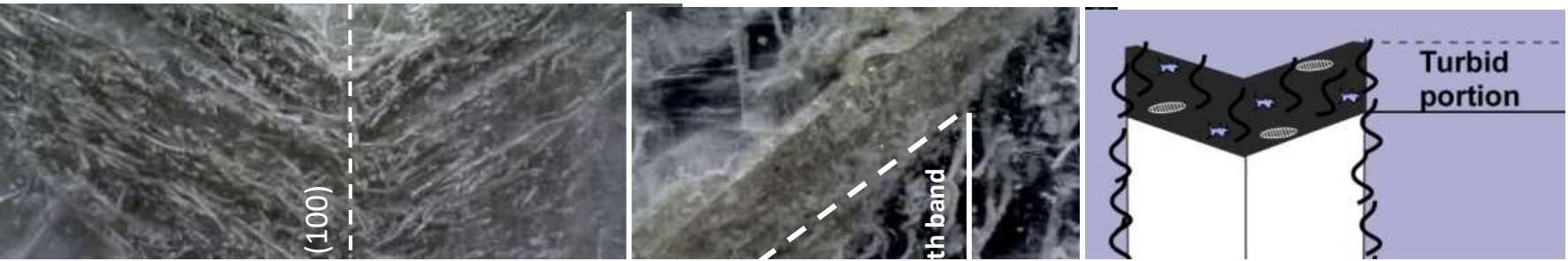


20  $\mu$ m

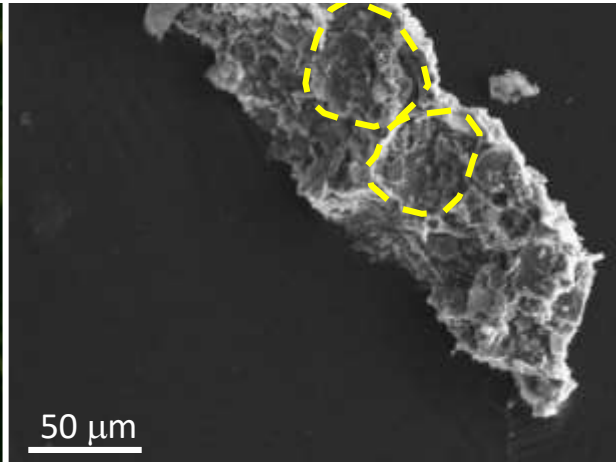
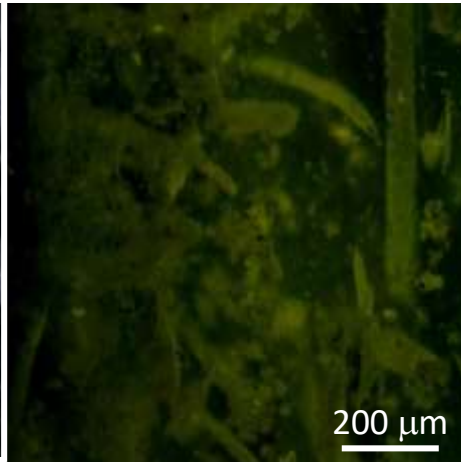
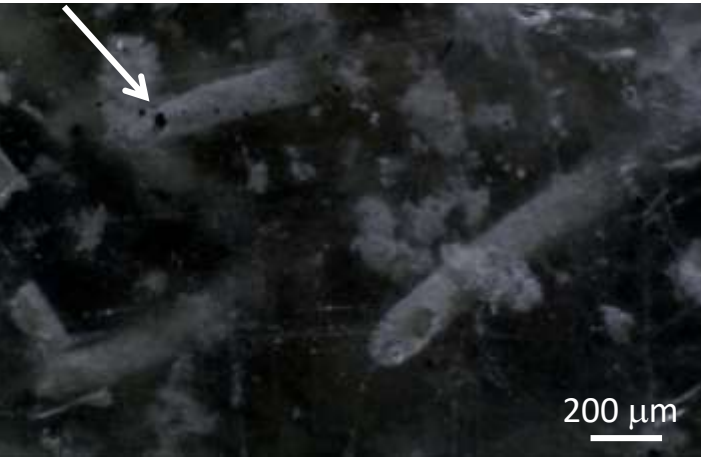
*Stefanodiscus*

Benthic components (in the reentrant angle and along the vertical growth bands)

# Filamentous microfossils

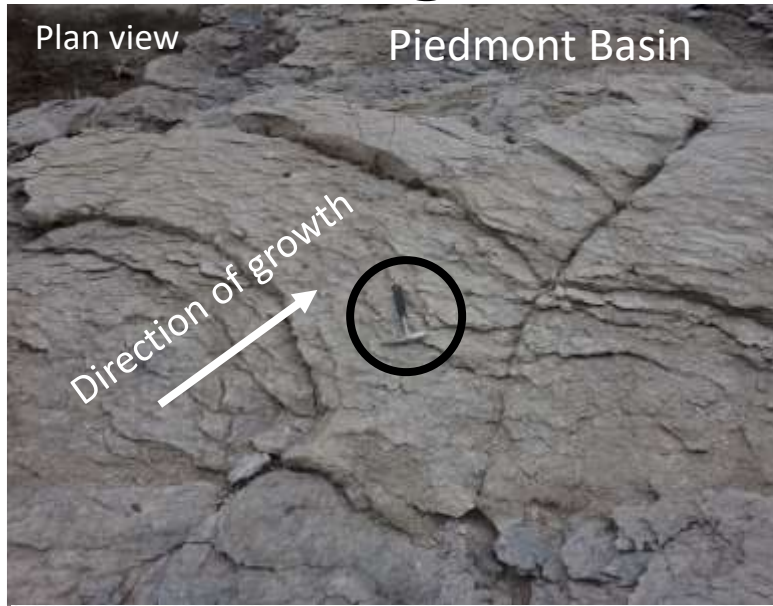


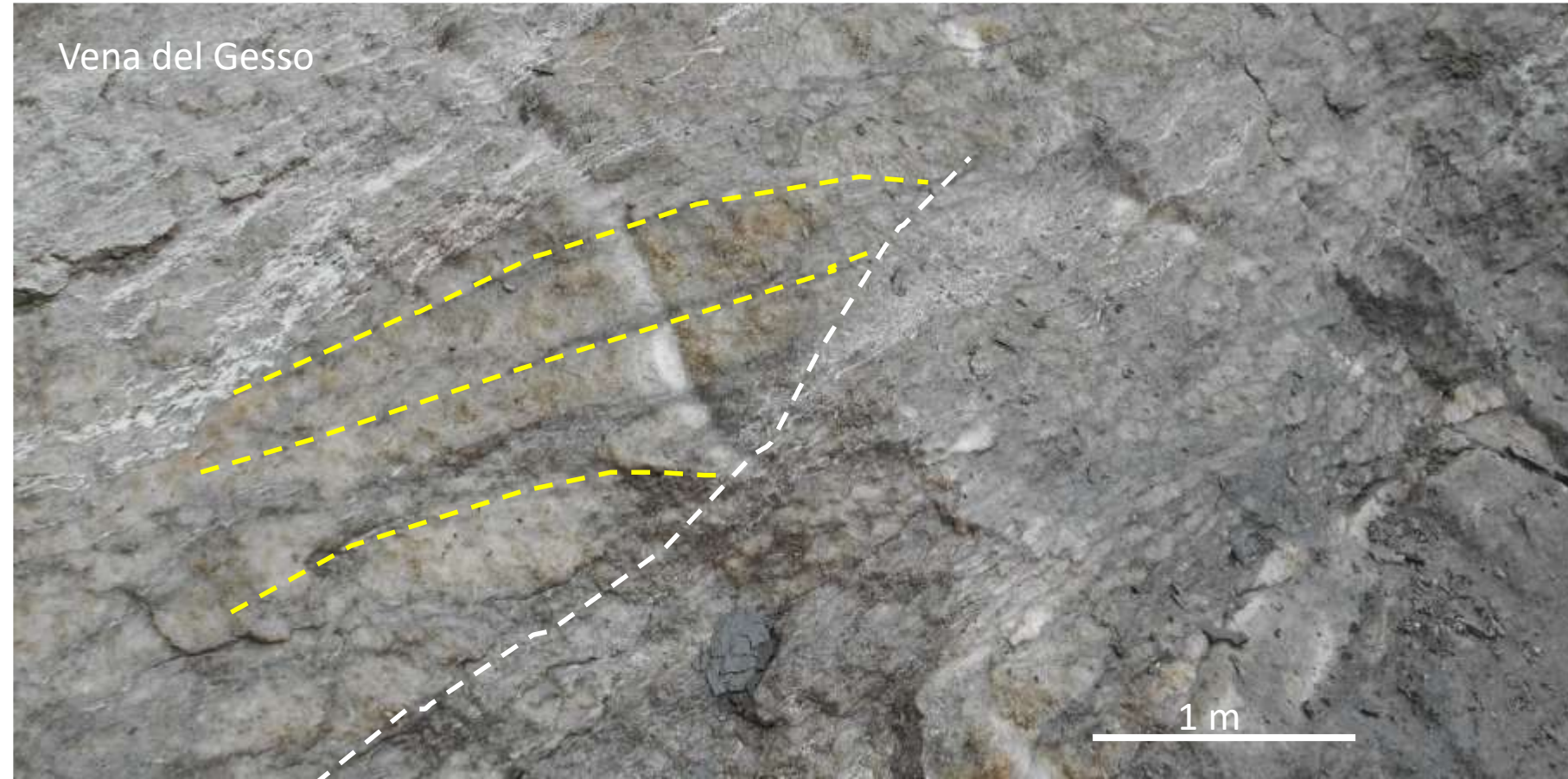
- ✓ Microbial sulfur cycling was an active process in the gypsum forming brines
- ✓ Deposition not necessarily in the photic zone



Cyanobacteria ?(Panieri et al., 2010) Sulfide-oxidizing bacteria? (Dela Pierre et al., 2015)

# Branching selenite



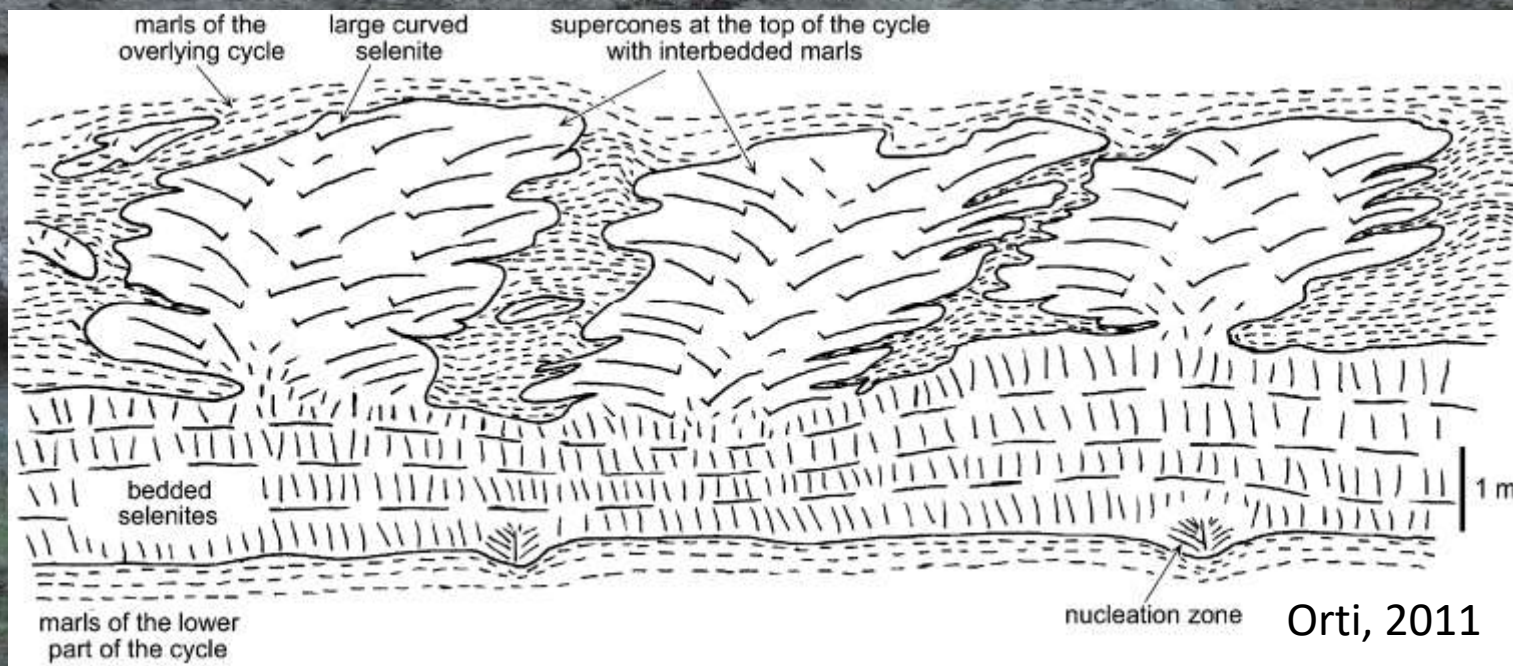
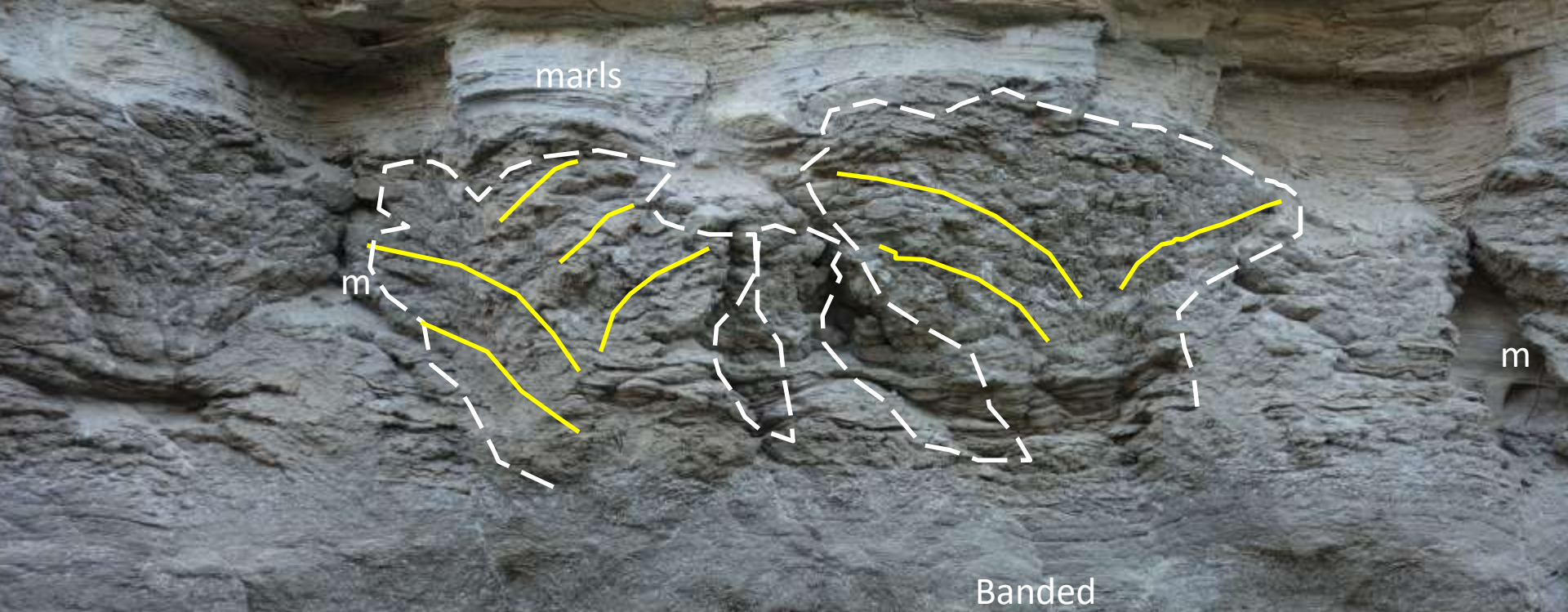




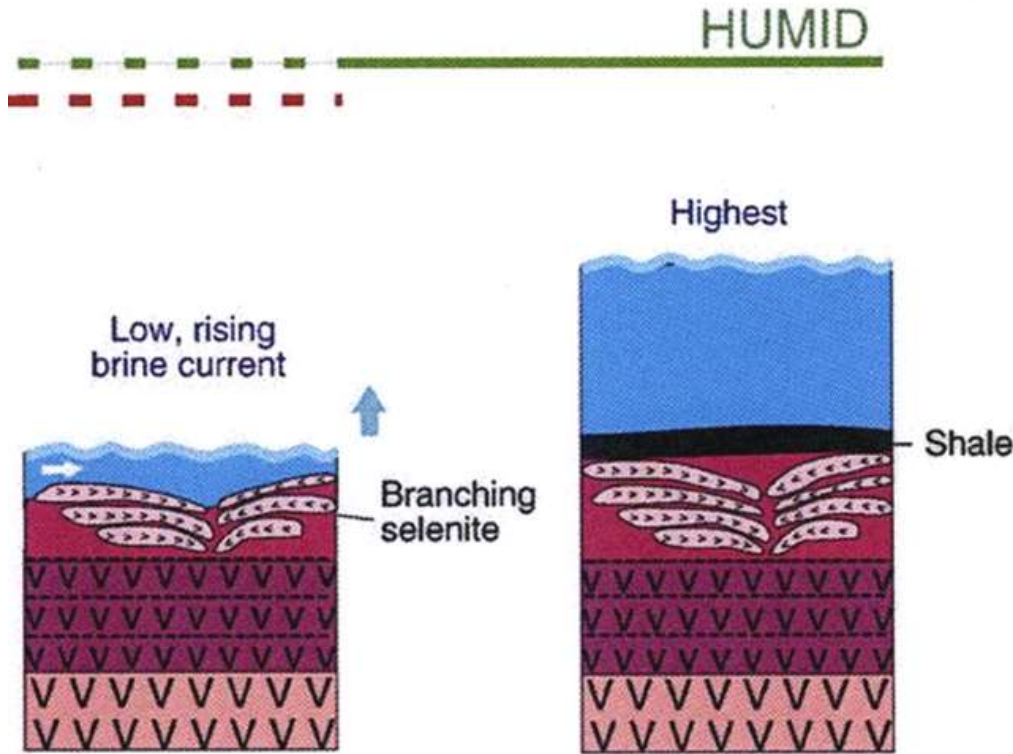
Branching selenite

Piedmont Basin

Laminar gypsum

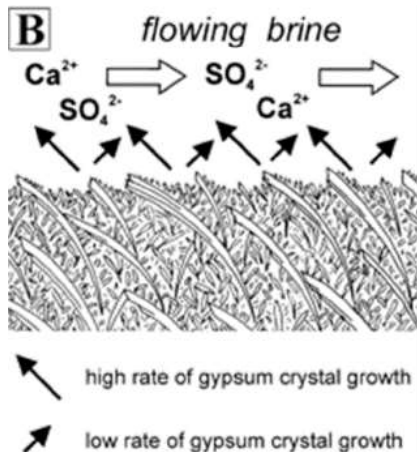


# Braching selenite: interpretation



Brine flows favouring the horizontal growth of the crystals, and a depressed pycnocline the force the strcture to grow laterally.

*Lugli et al., 2010*



Brine currents, with preferential growth of the crystals upstream, i.e towards the ions

*Babel, 2002*

# Laminar gypsum ("Balatino")

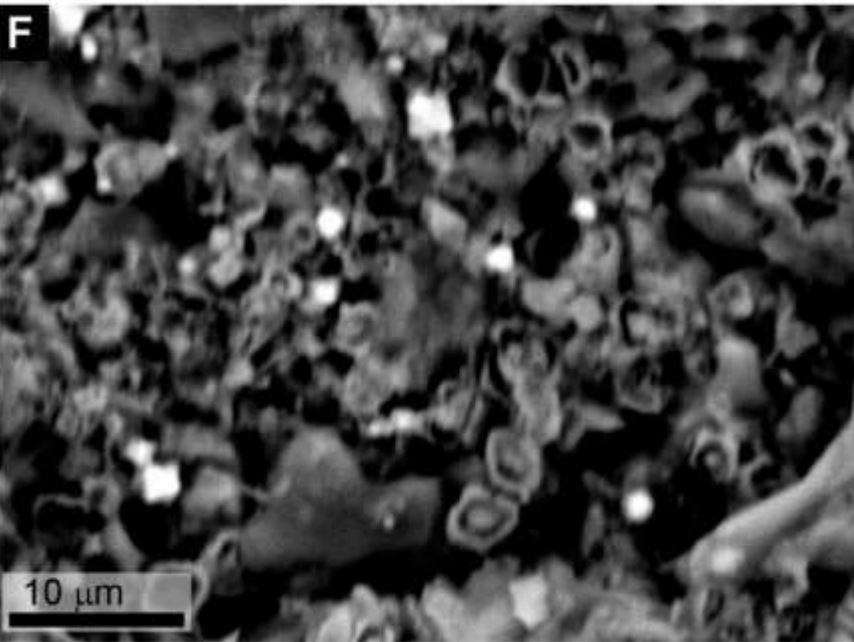
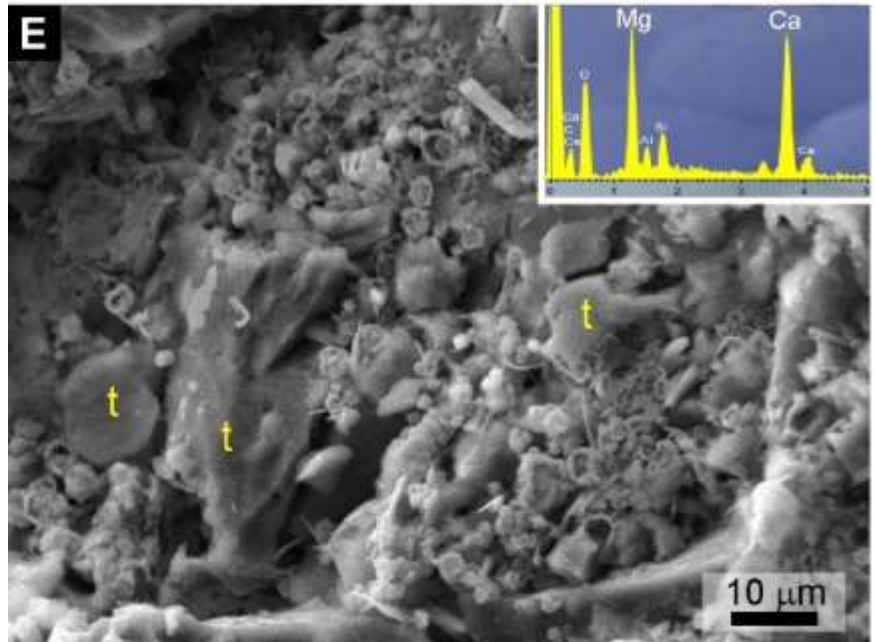
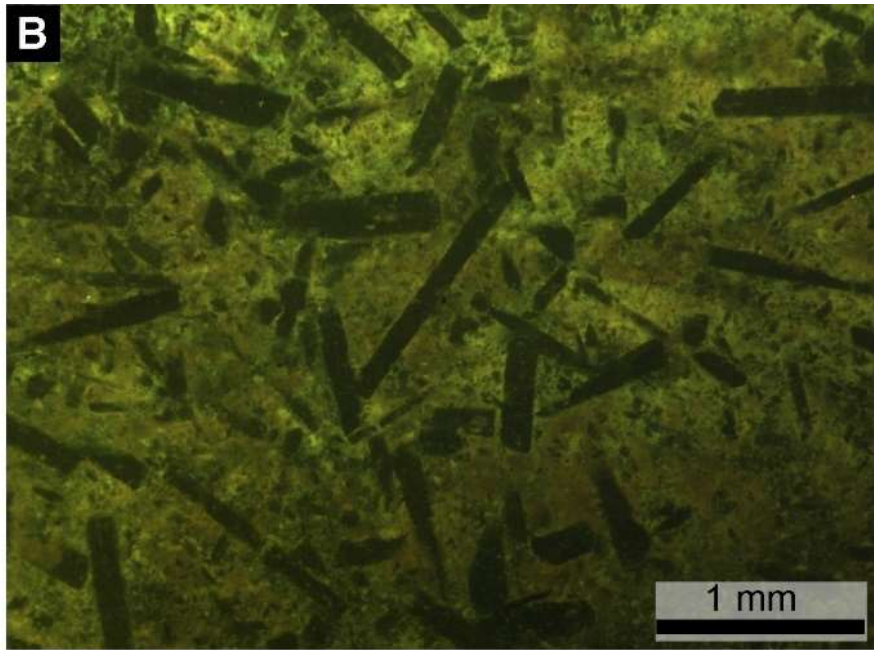
Laminated gypsum (cumulites);  
Messinian



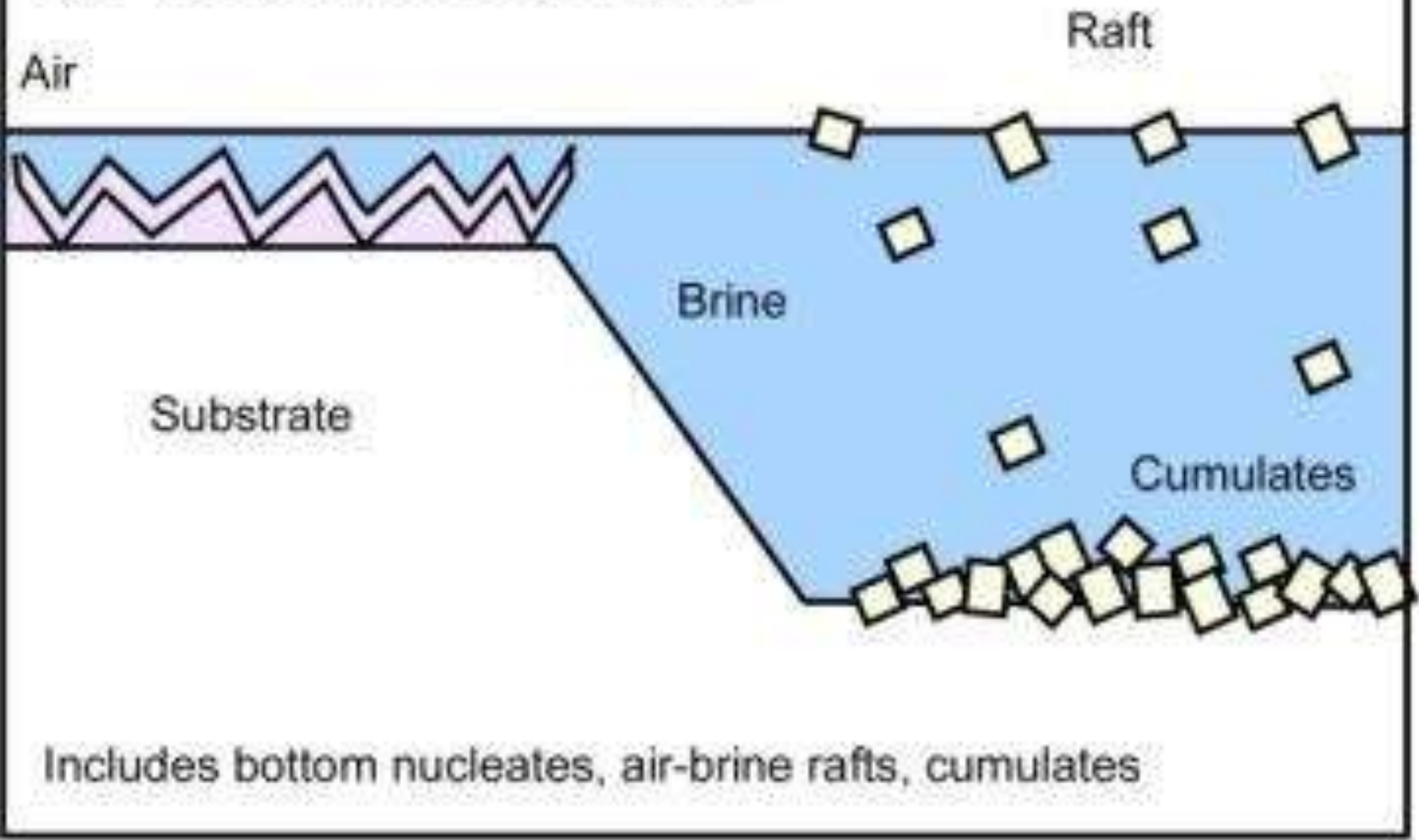
Ancient evaporites



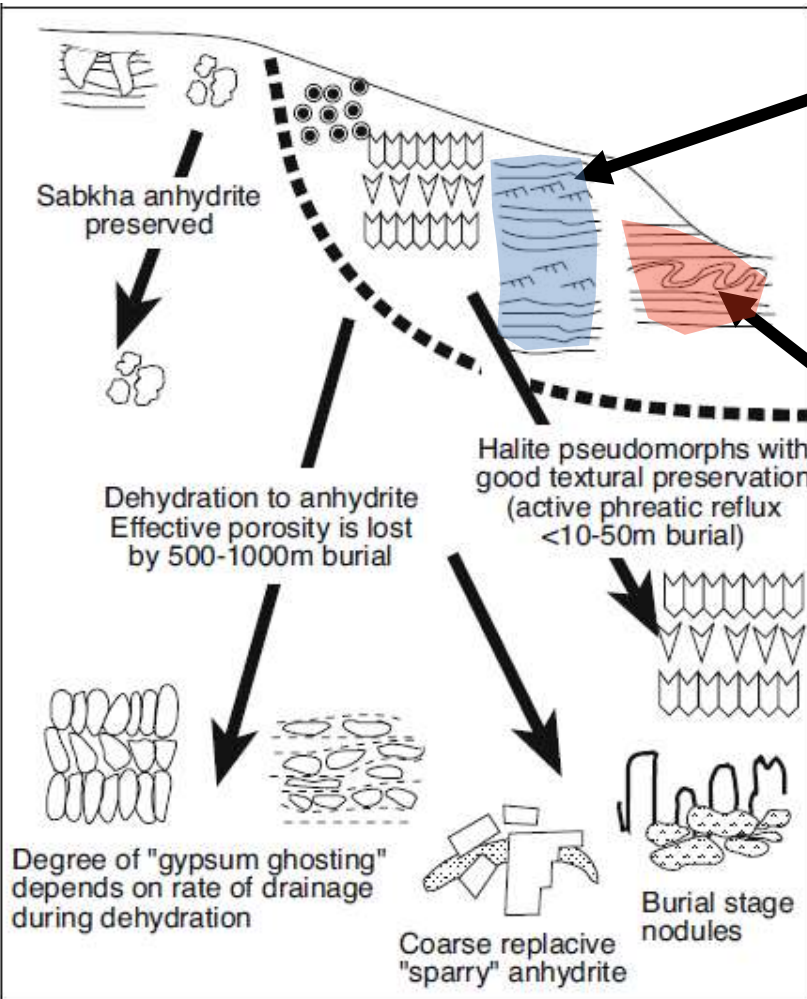
# Laminar gypsum ("Balatino")



# 1) *Primary precipitates*



# CLASTIC GYPSUM: SHALLOW & DEEP WATER



Shallow water: clastic sedimentary structures (Cross bedding, ripples ecc..)

Deep water: Mass flows, Turbidites, Slumps and Slides

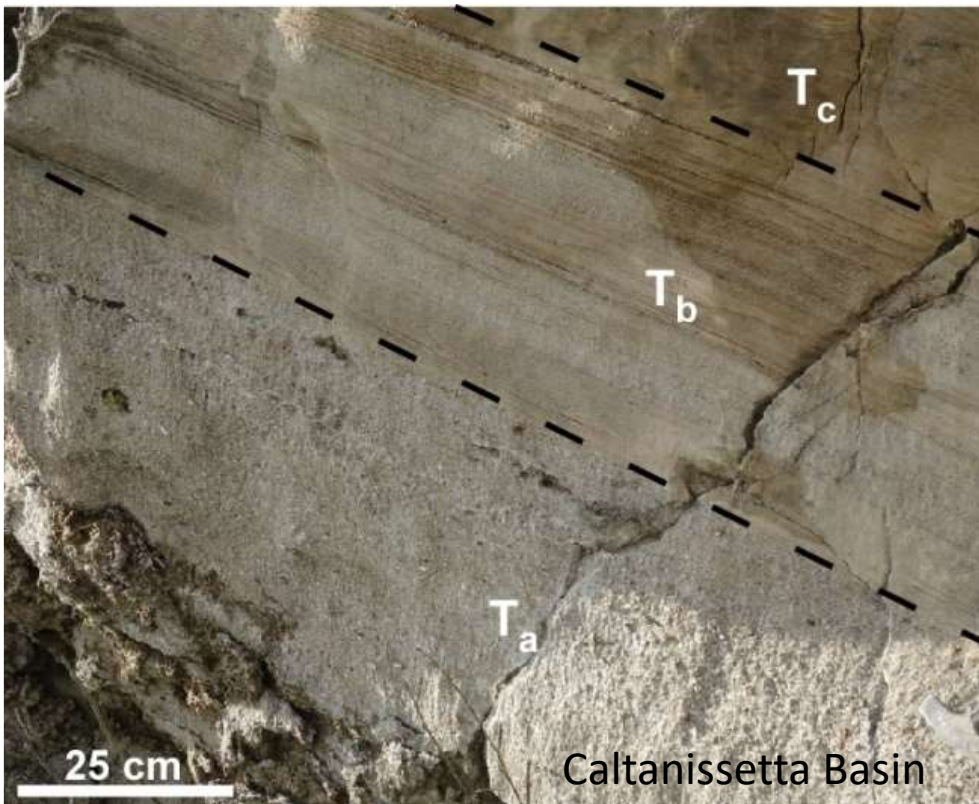
# Clastic evaporites (shallow water)



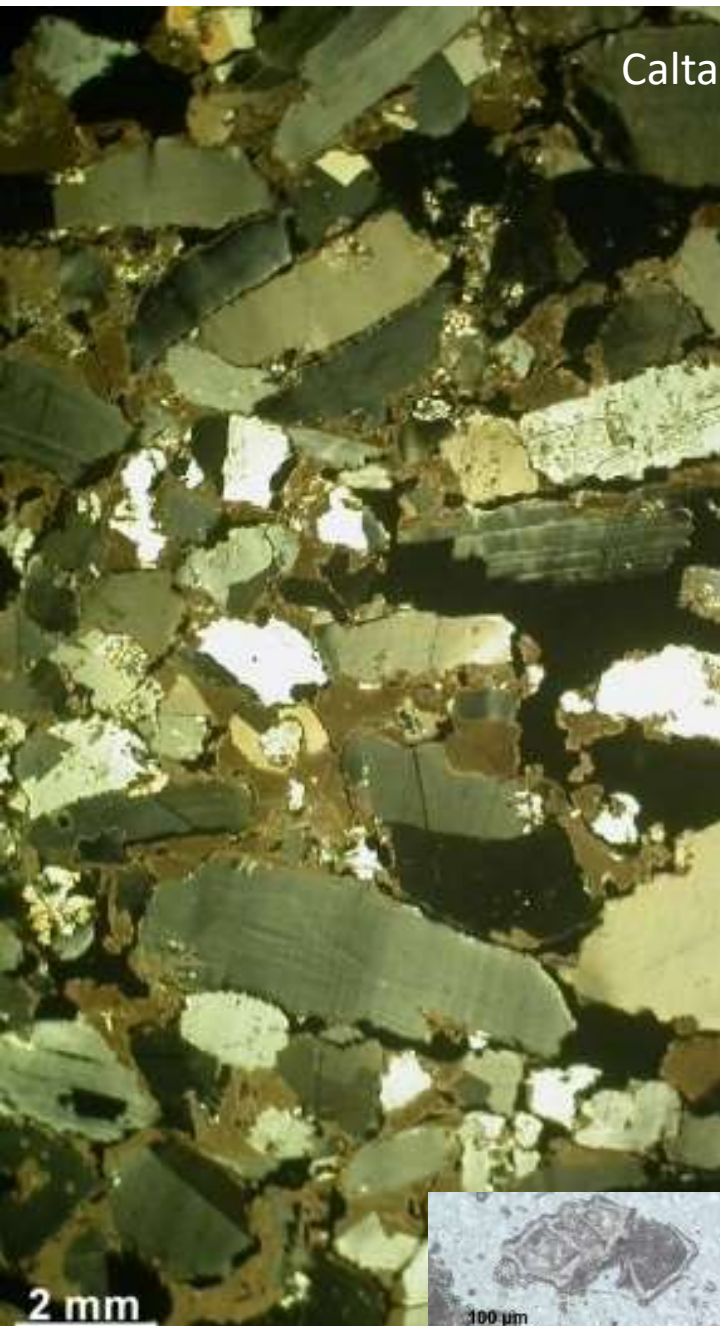
Piedmont Basin



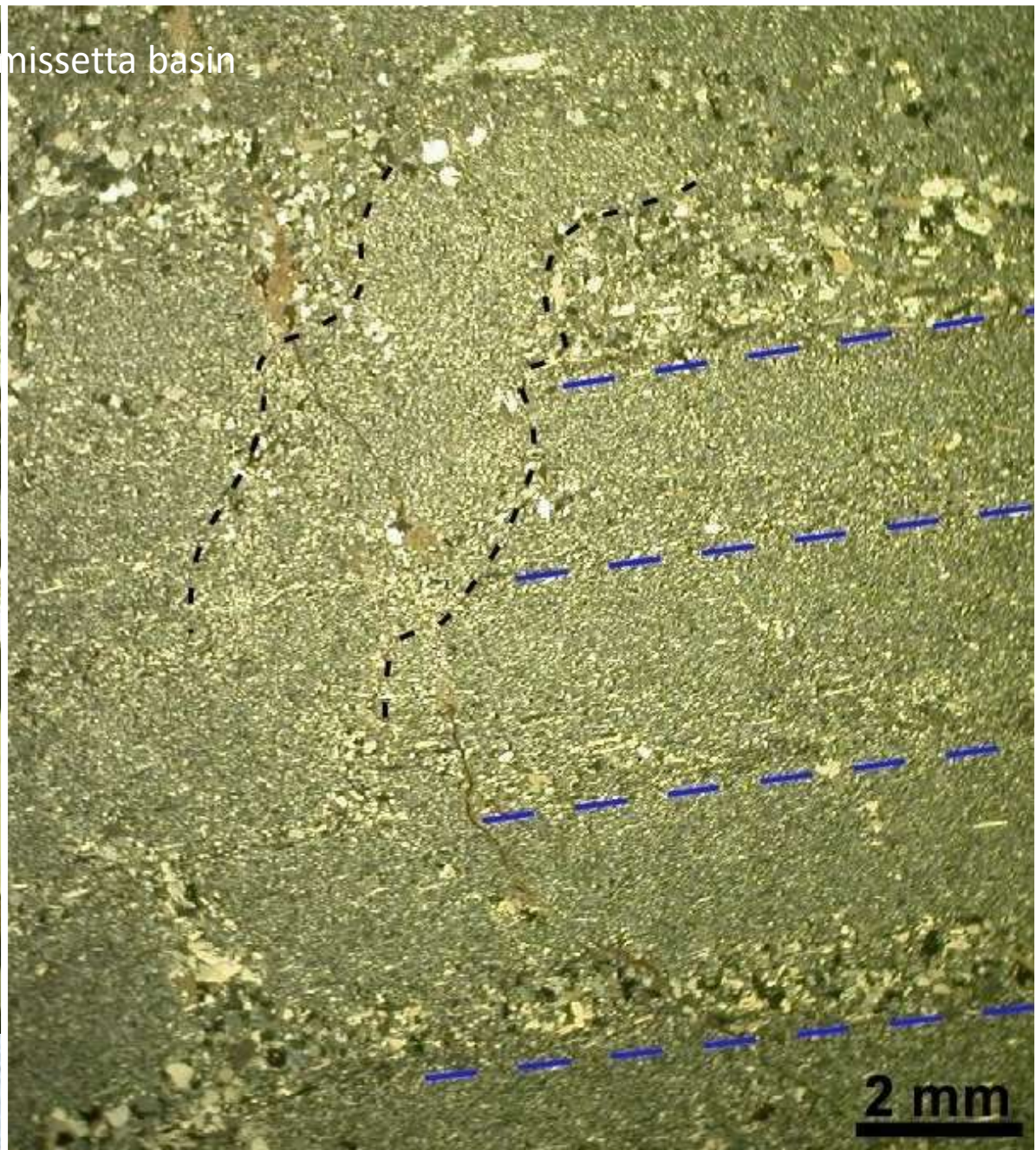
# Clastic evaporites ("deep" water)



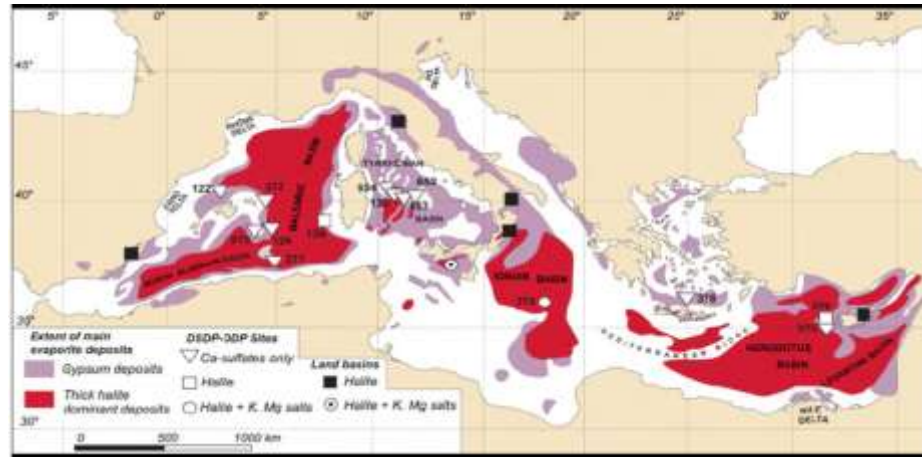
# Clastic evaporites



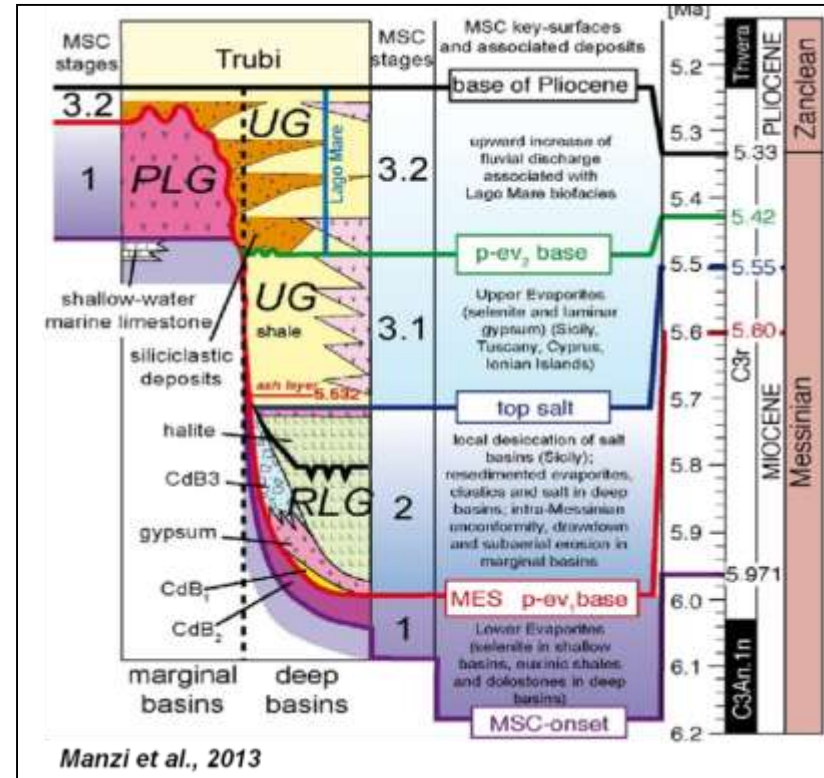
Caltanissetta basin



# Ancient evaporites (MSC)



Rouchy and Caruso 2006



Manzi et al., 2013

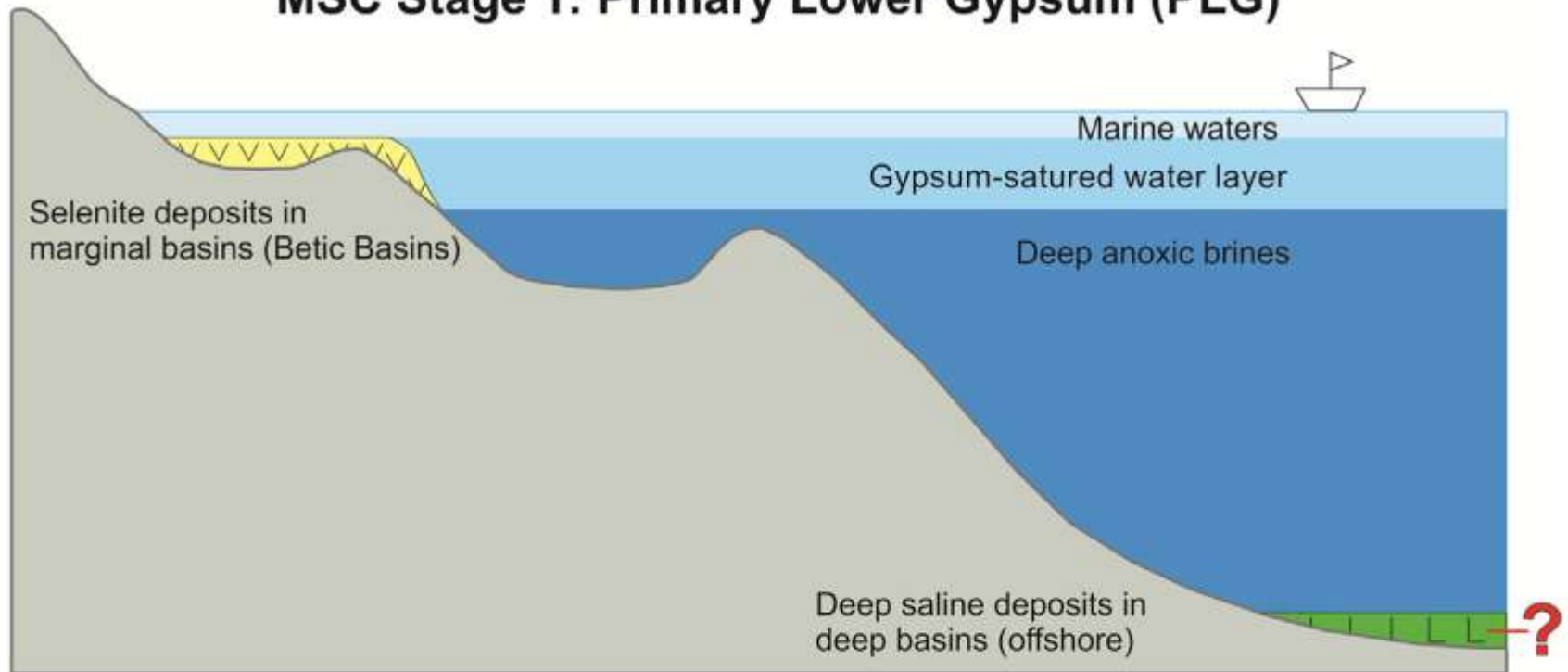
Interpretation of ancient marine evaporite deposits is difficult!

- No (few) clues to estimate water depth
- No modern analogues for very large and (relatively) deep basins (Salt giants)

This is particularly true for the wide array of Messinian evaporite facies (mostly gypsum and halite) of the Mediterranean!

# MSC gypsum deposits: the PLG unit (and lateral equivalents)

## MSC Stage 1: Primary Lower Gypsum (PLG)



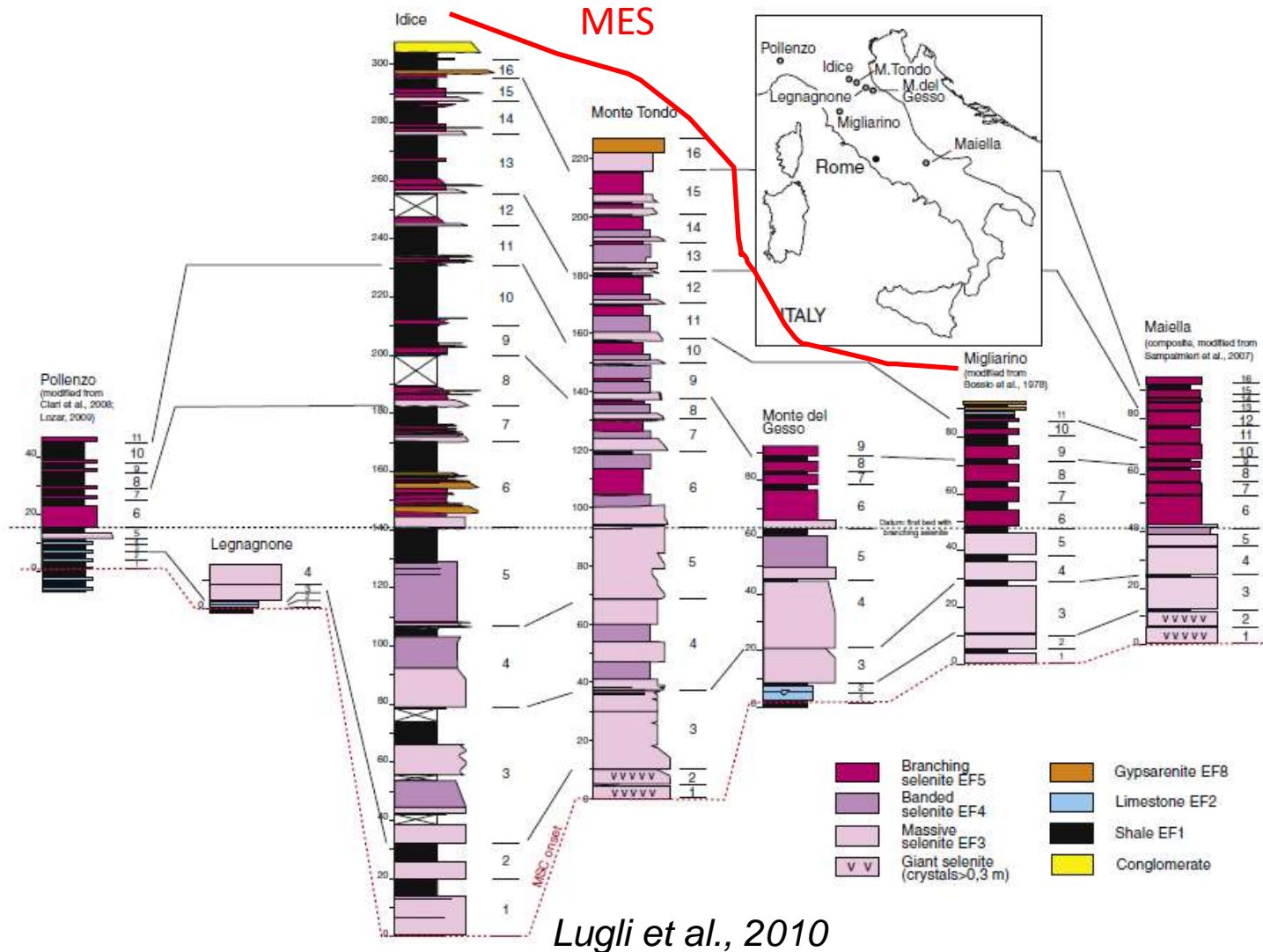
Garcia Veigas et al., 2018

Primary gypsum (bottom grown and laminar) in peripheral marginal basins within a gypsum saturated layer.

In intermediate to deep basinal areas, no gypsum but organic-rich shales and possibly halite (Melijson et al., 2019)



# PRIMARY LOWER GYPSUM UNIT (PLG) (5.97-5.60 Ma)



# Messinian Lower Gypsum deposits

Thickness: 200 m

In < 300 kyr



*Vena del gesso (Apennines, Italy)*

*Vena del gesso (Apennines, Italy)*



# Vena del Gesso

*Brisighella quarry (Northern Apennines)*



Messinian erosional surface (MES)

Pliocene

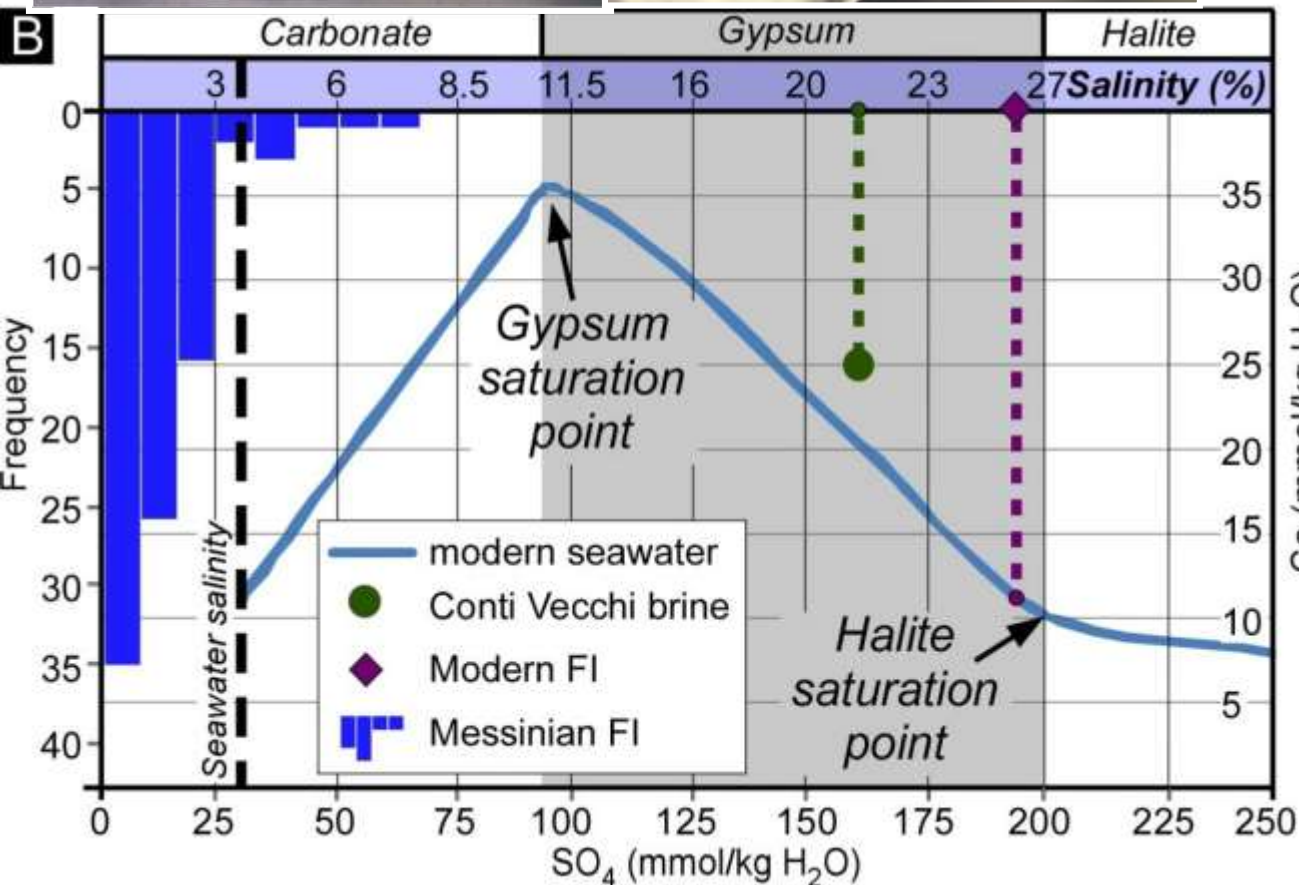
Lago Mare

Gypsum (PLG)

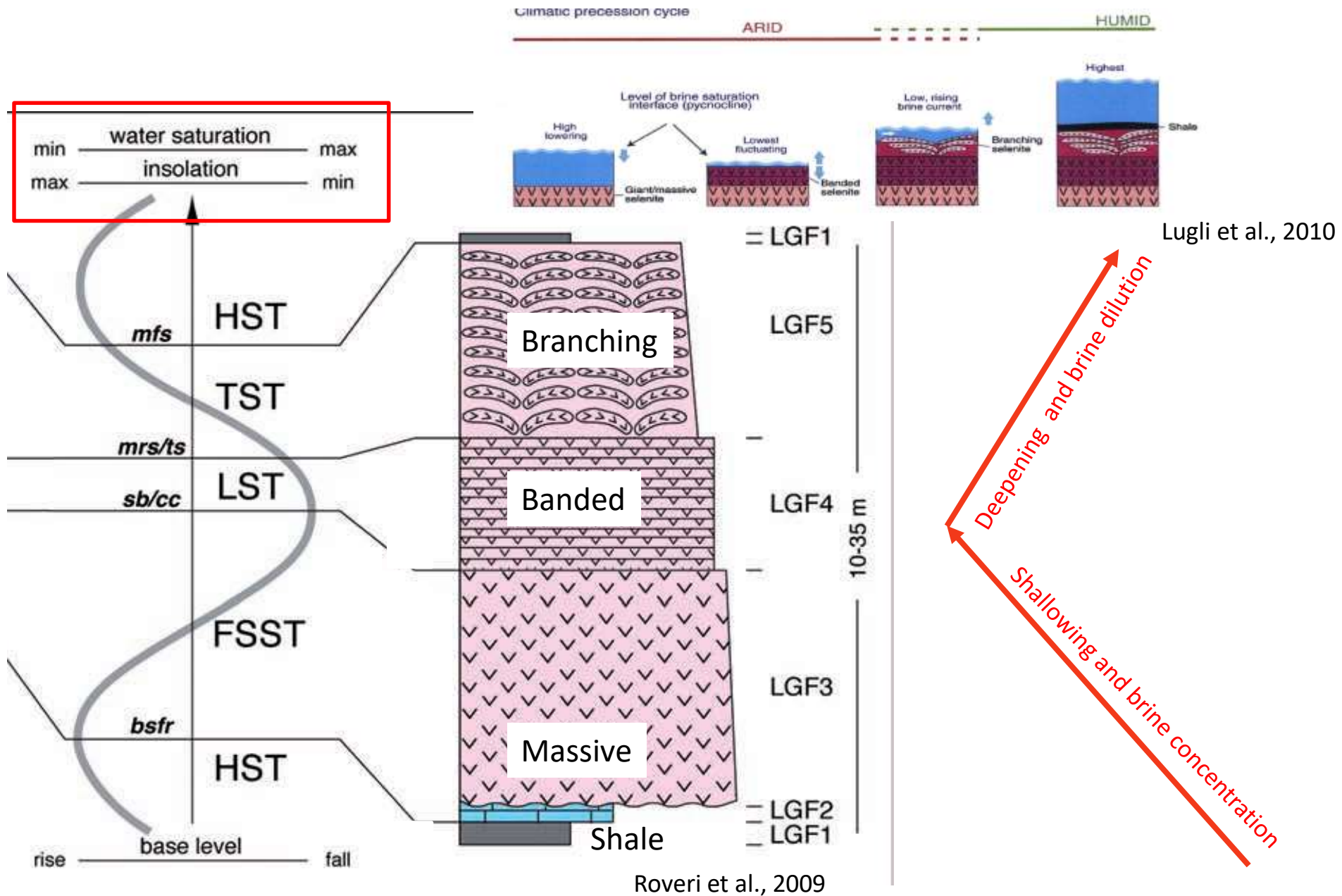
# Salinity of the parent brines of massive selenite

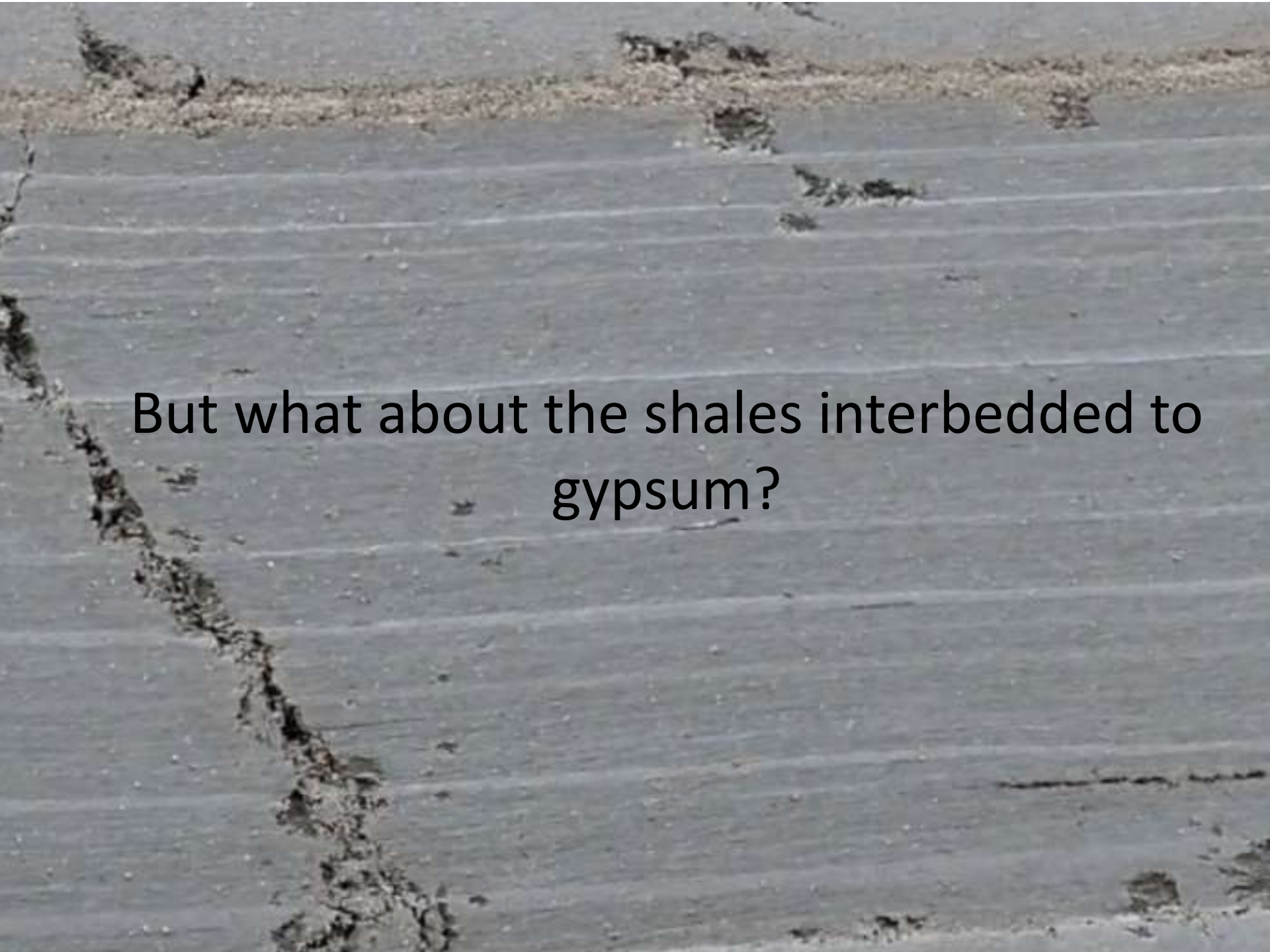


Lower salinity than not evaporated seawater (below the gypsum saturation point)



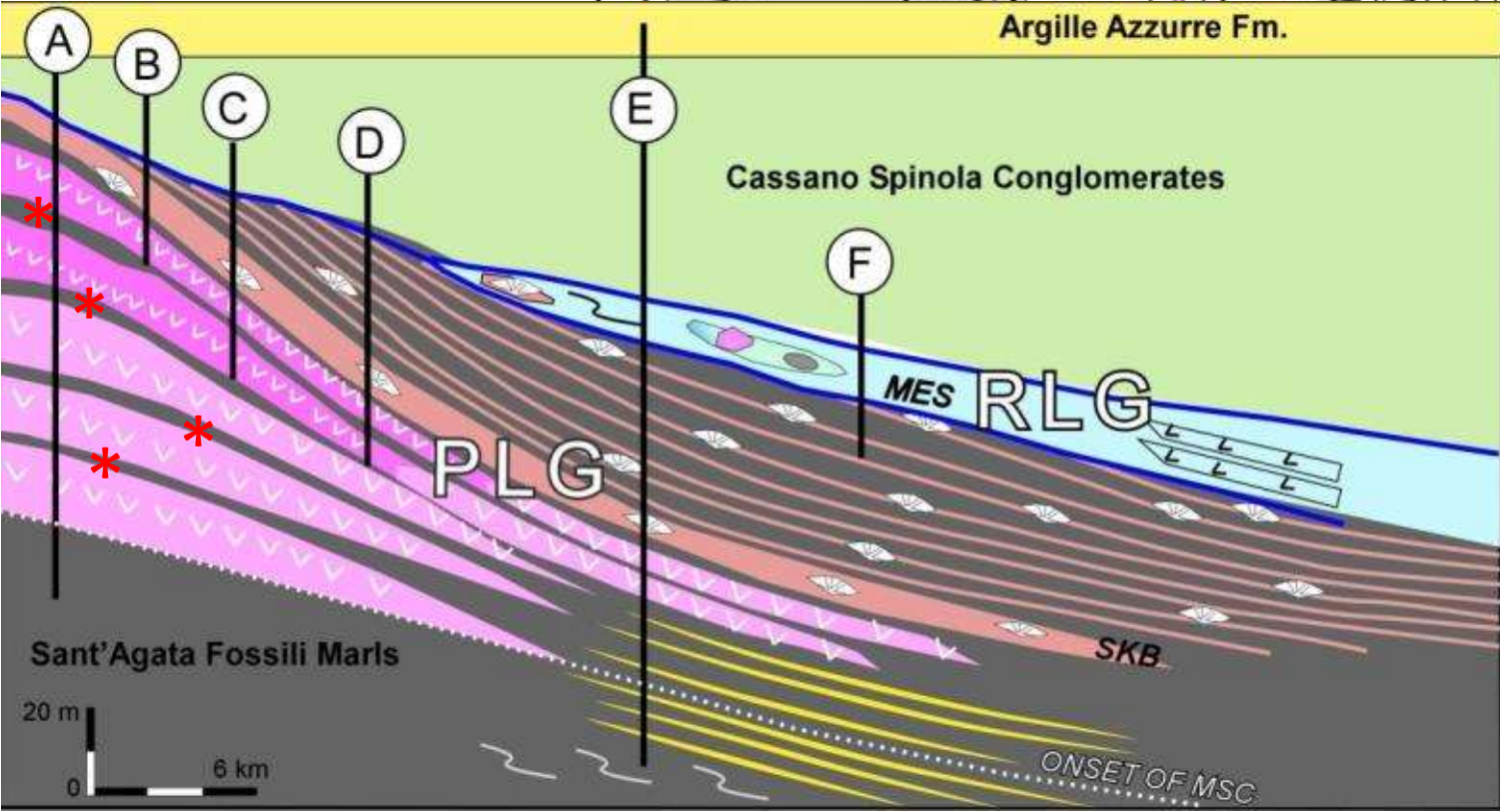
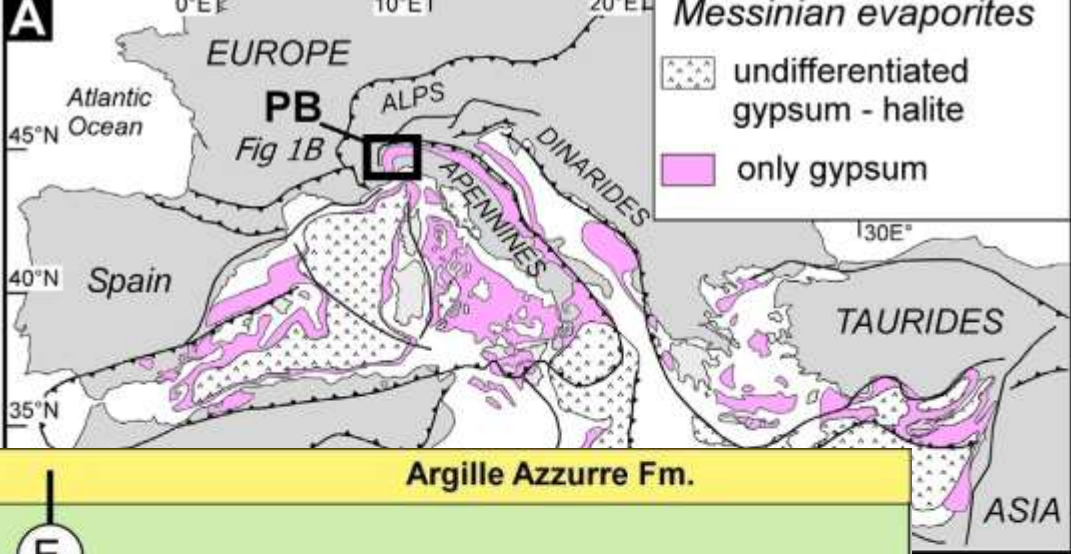
# Ideal facies cycle in the PLG unit (Vena del Gesso)



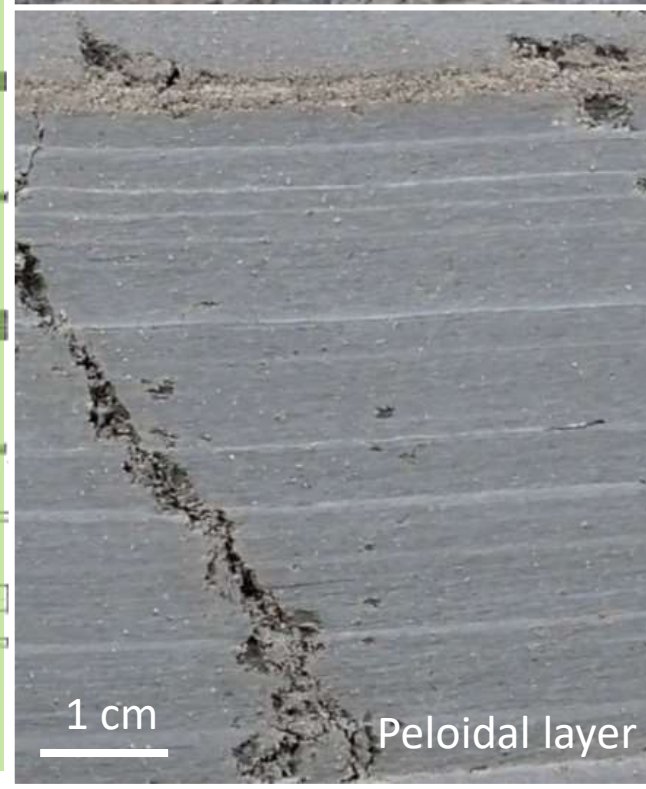
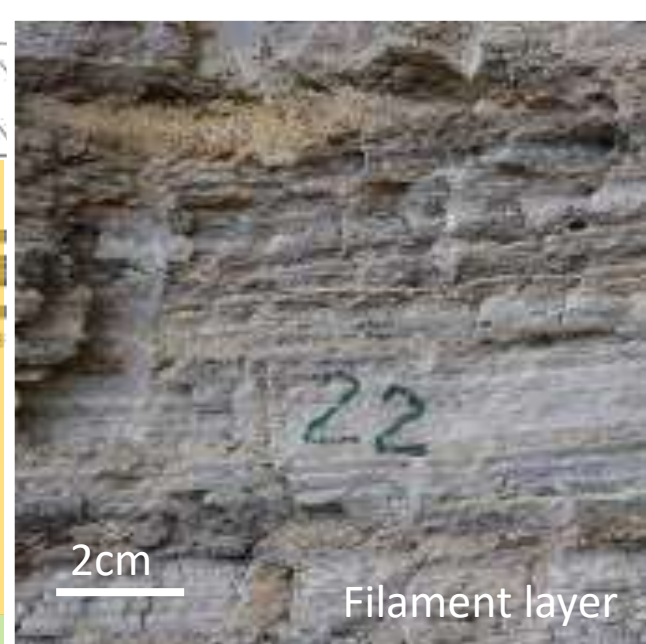
A photograph of a geological outcrop showing alternating layers of dark shale and lighter-colored gypsum, illustrating interbedding. The layers are roughly horizontal and show some fracturing. The text is overlaid on the center of the image.

But what about the shales interbedded to  
gypsum?

# Piedmont Basin

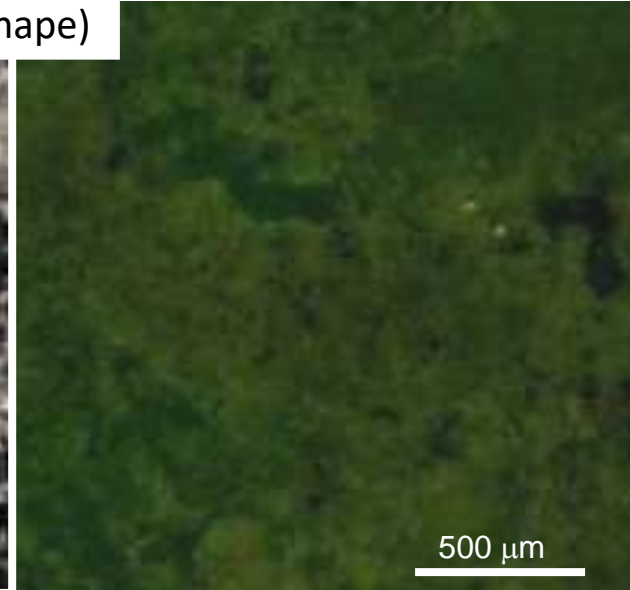
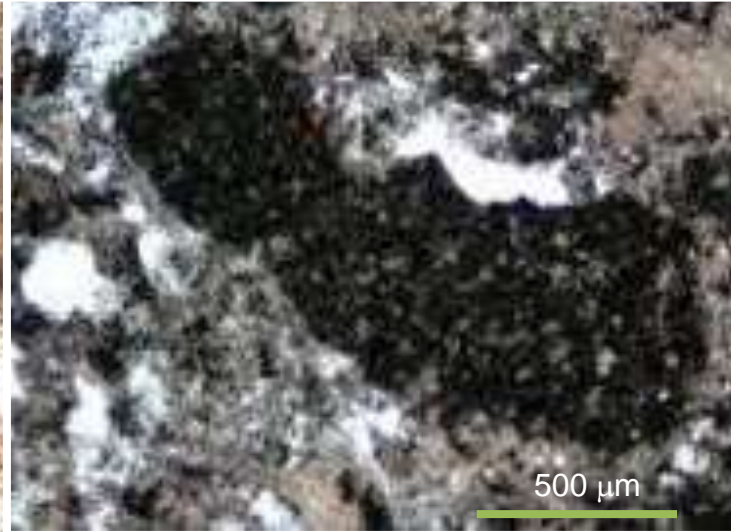
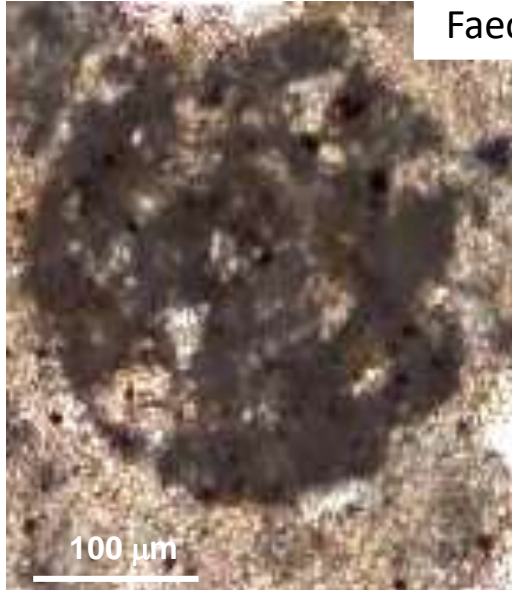




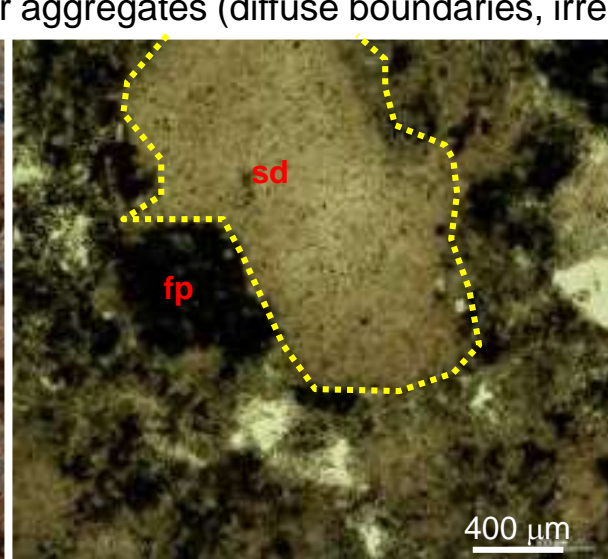
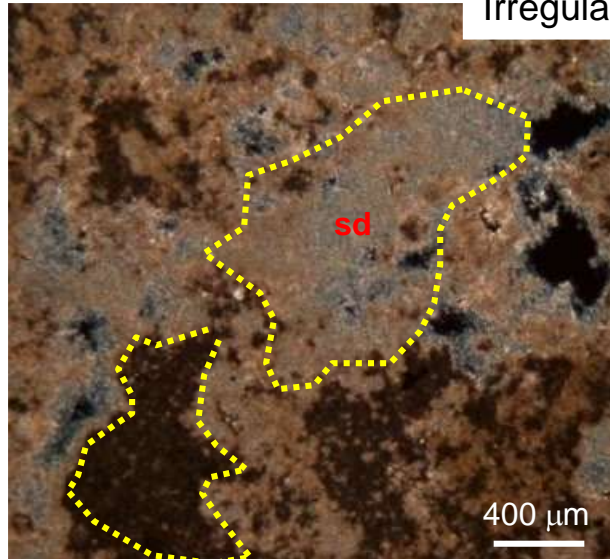


# Peloidal layers

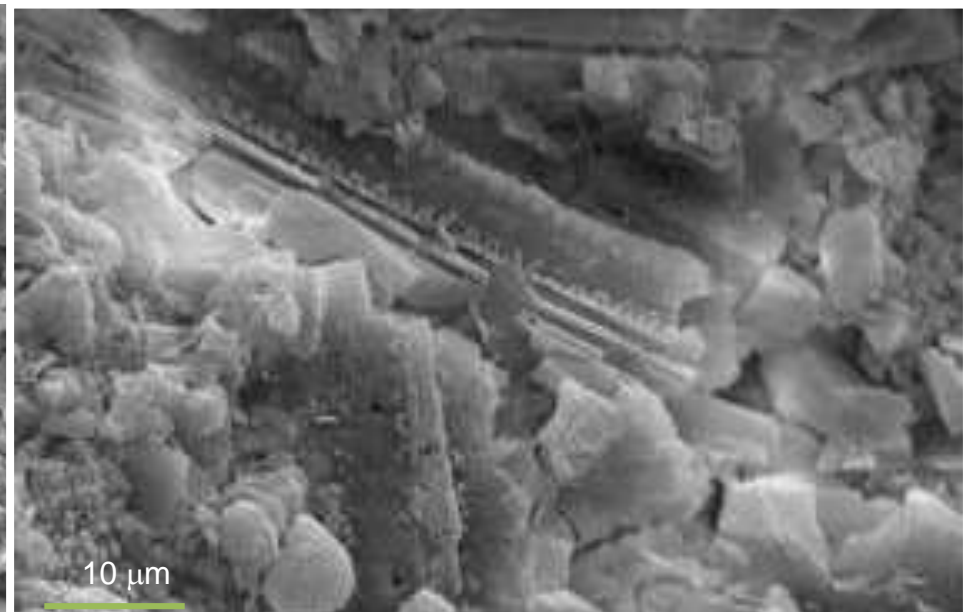
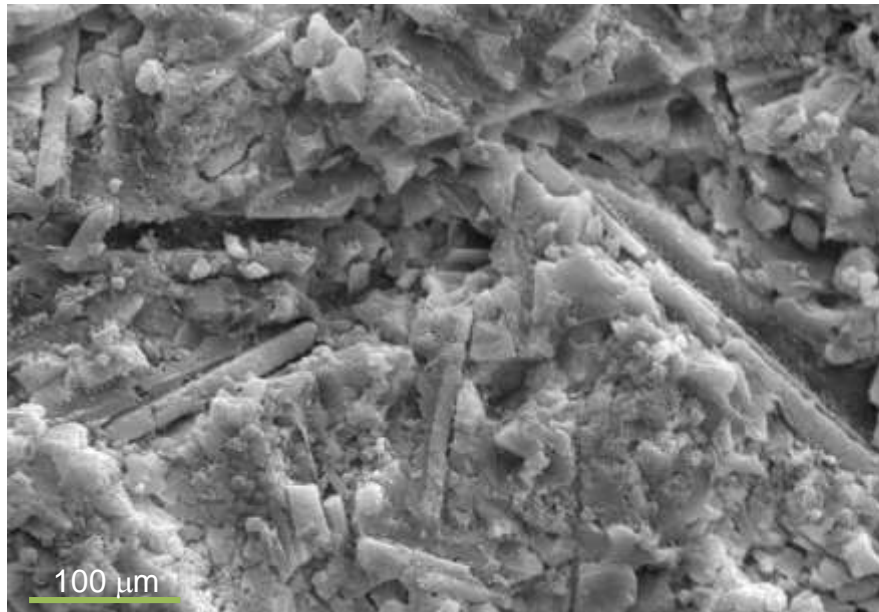
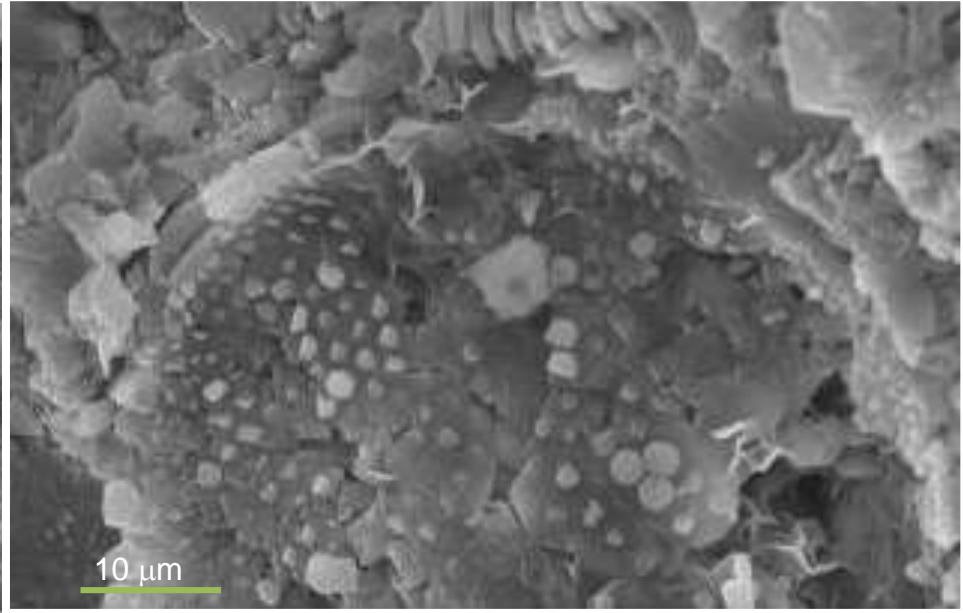
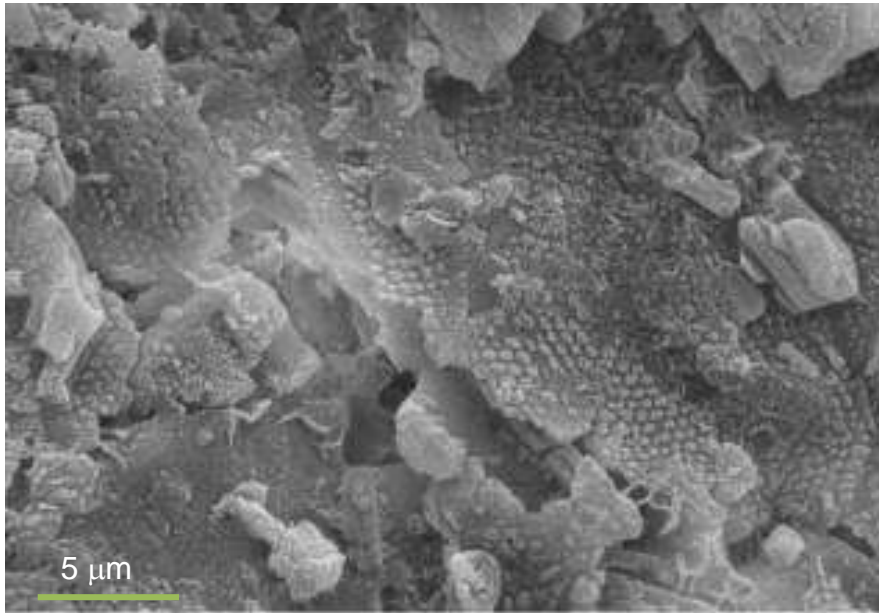
Faecal pellets (sharp boundaries and regular shape)



Irregular aggregates (diffuse boundaries, irregular shape)



Ghost of diatoms within the aggregates. The skeleton is often dissolved and replicated by dolomite crystals

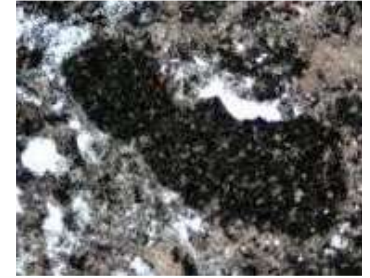
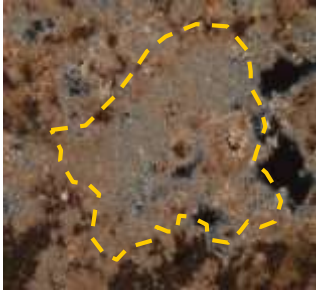


# Peloidal layers

**Irregular aggregates  
= marine snow**



**Faecal pellets**



Messinian aggregate

Present-day marine snow

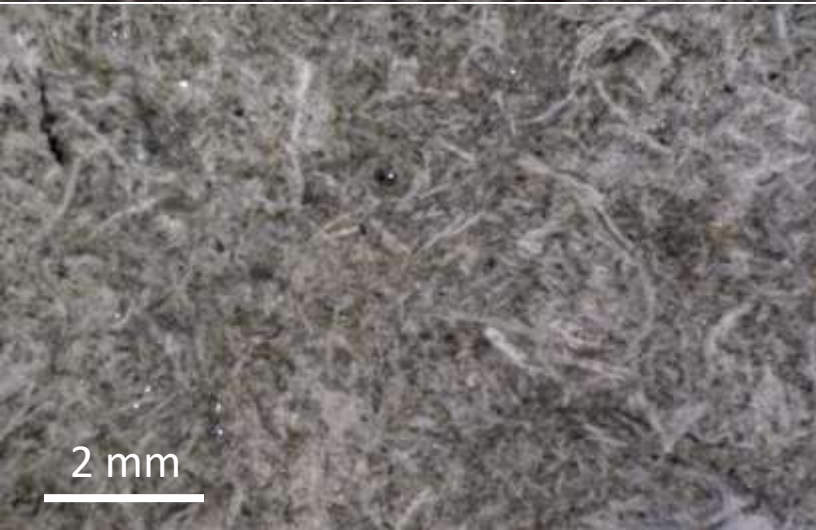
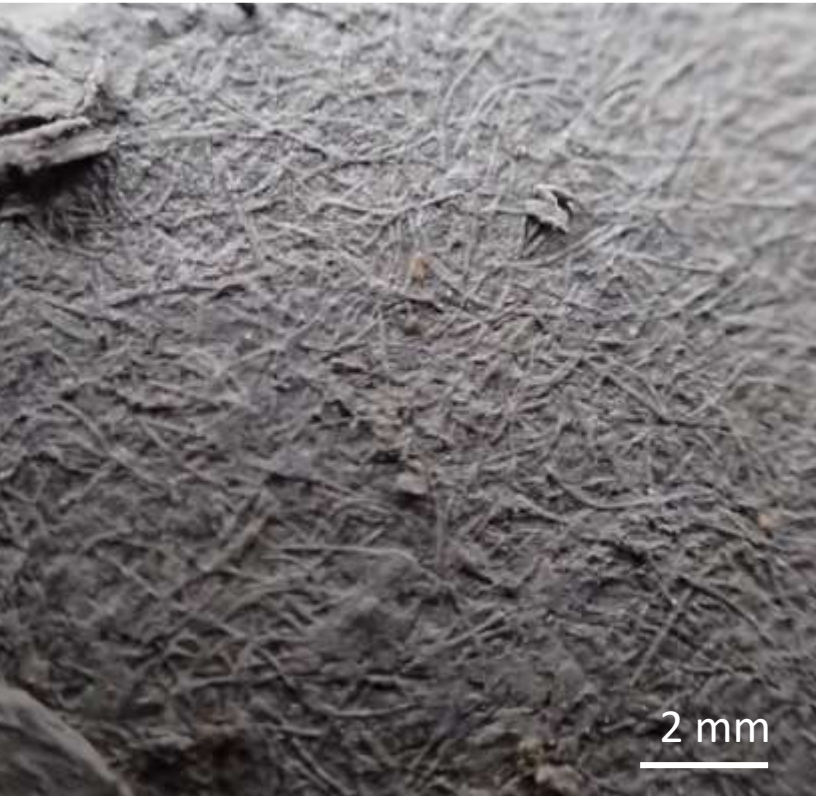
Present day faecal pellets

Messinian peloids

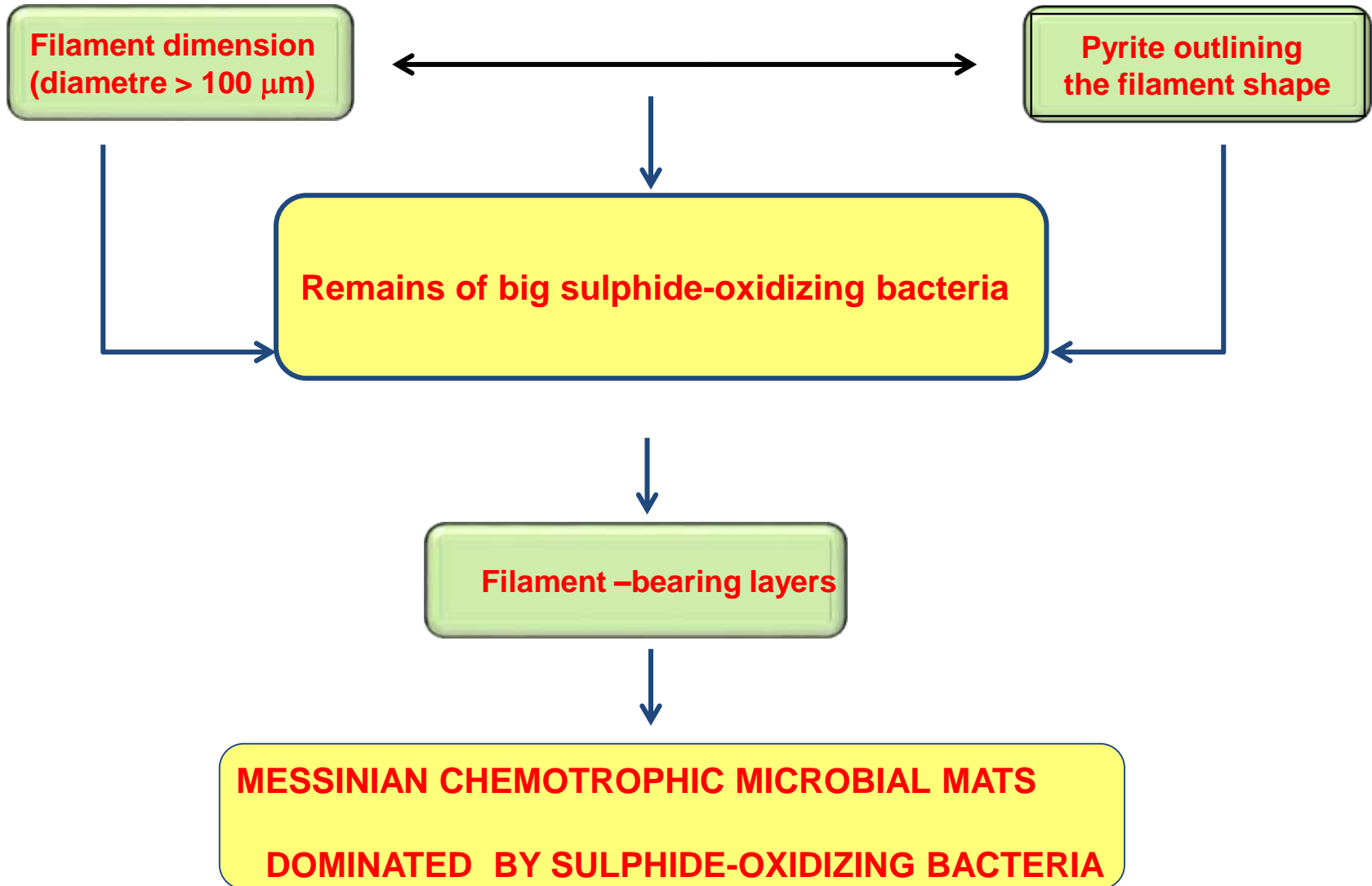
**PRESERVATION OF FLOCCULENT LAYERS**

- ✓ Episodes of high organic productivity in the upper water column
- ✓ High rate of sulphate reduction and production of hydrogen sulphide

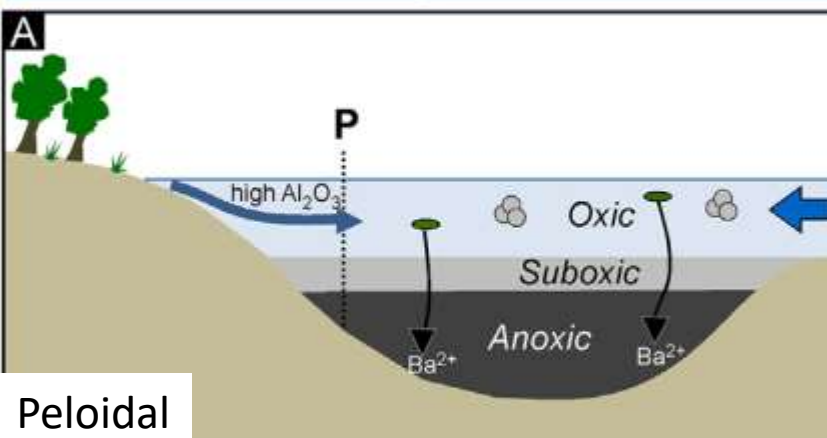
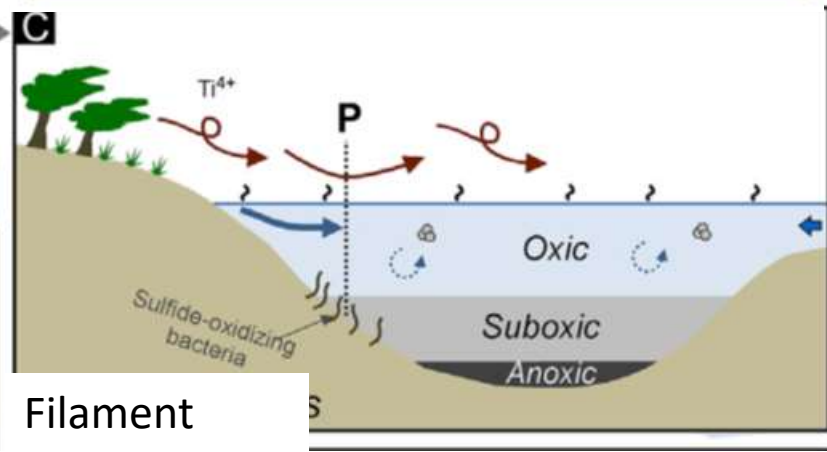
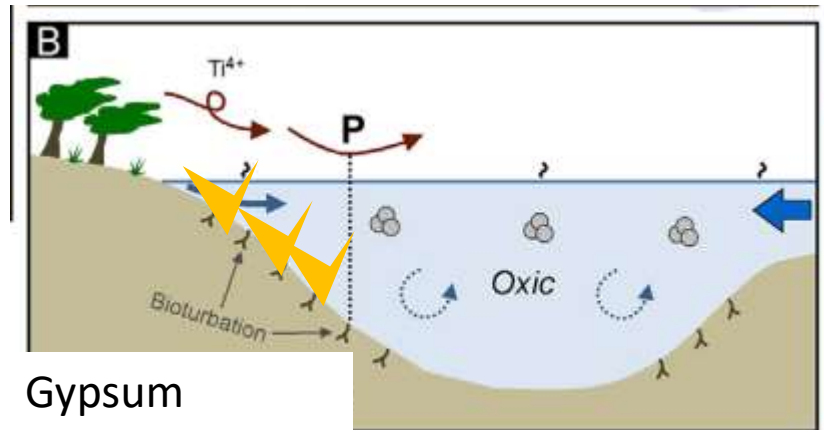
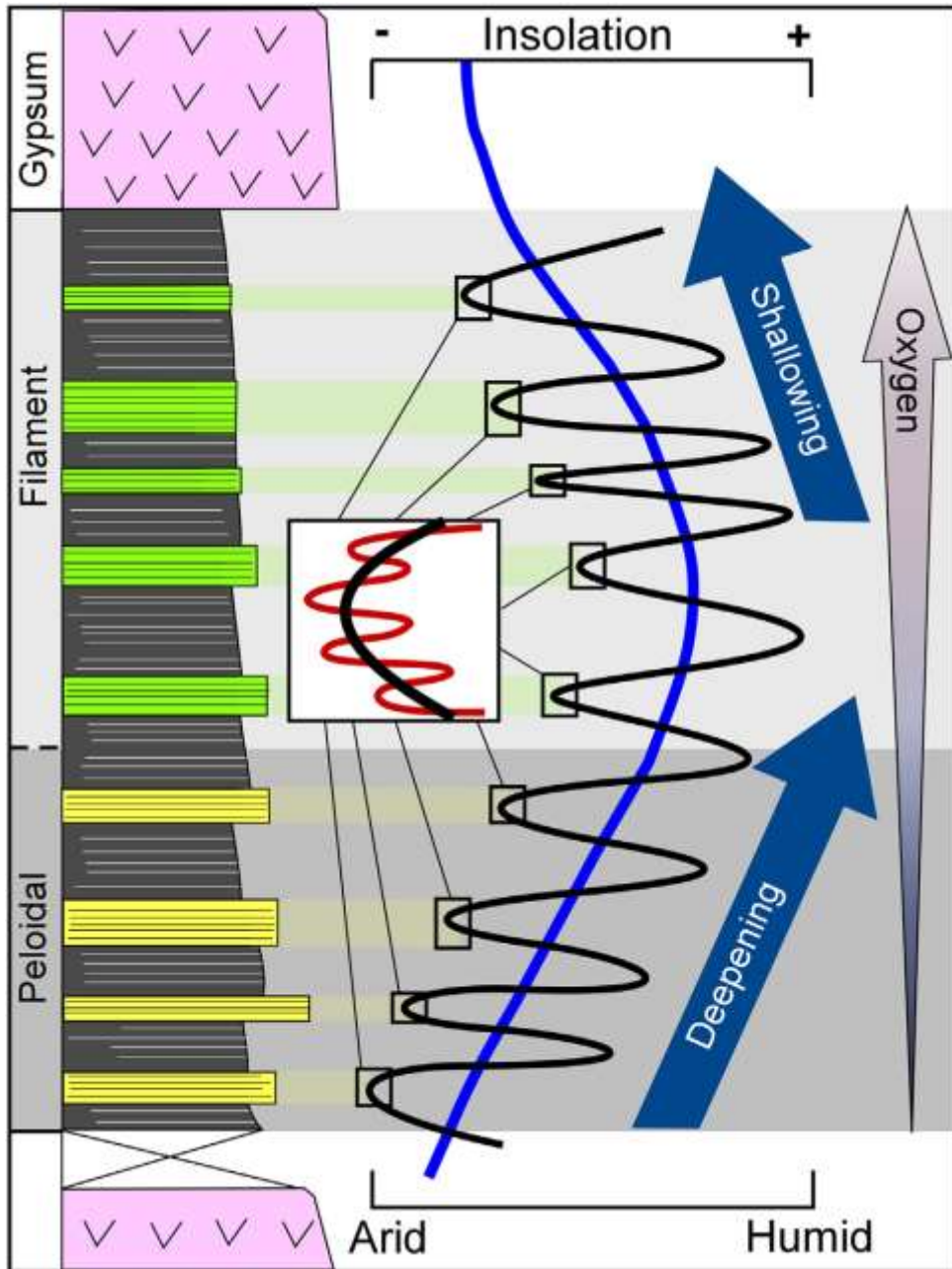
# Filament-bearing layers



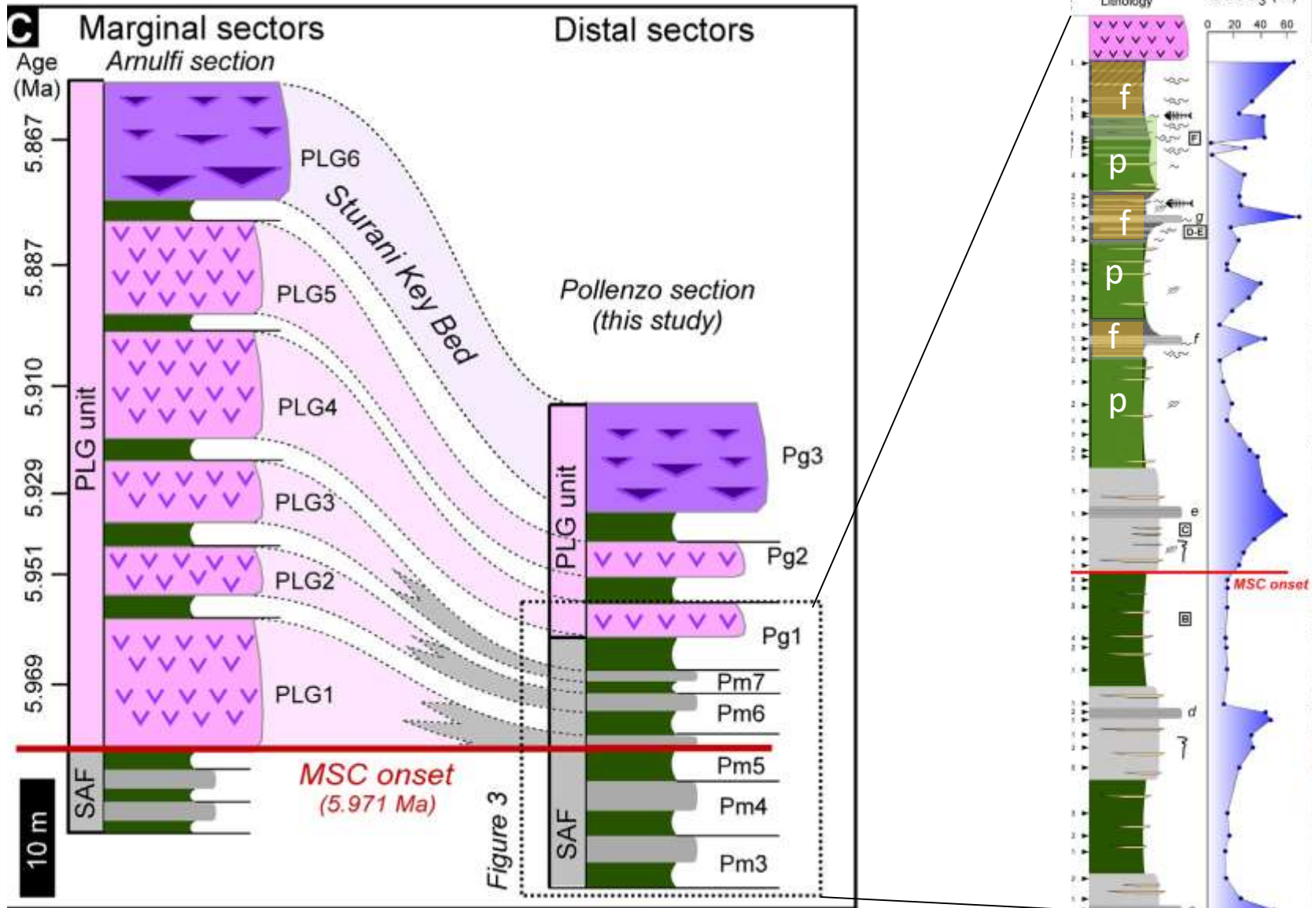
# Interpretation



# Ideal facies cycle in the shales



# Similar stacking pattern in the deep water equivalents of marginal gypsum (Pollenzo section)



But now, let' move to organic geochemistry.....



Grazie per l'attenzione!

